

=> fil reg

FILE 'REGISTRY' ENTERED AT 18:55:04 ON 01 JUN 2007  
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STRUCTURE FILE UPDATES: 31 MAY 2007 HIGHEST RN 936320-32-0  
 DICTIONARY FILE UPDATES: 31 MAY 2007 HIGHEST RN 936320-32-0

New CAS Information Use Policies, enter HELP USAGETERMS for details.

TSCA INFORMATION NOW CURRENT THROUGH December 2, 2006

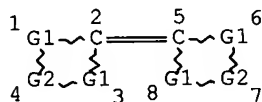
Please note that search-term pricing does apply when  
 conducting SmartSELECT searches.

REGISTRY includes numerically searchable data for experimental and  
 predicted properties as well as tags indicating availability of  
 experimental property data in the original document. For information  
 on property searching in REGISTRY, refer to:

<http://www.cas.org/support/stngen/stndoc/properties.html>

=> d que stat 18

L3 STR



VAR G1=S/O/SE/TE

REP G2=(1-5) C

NODE ATTRIBUTES:

DEFAULT MLEVEL IS ATOM

DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:

RING(S) ARE ISOLATED OR EMBEDDED

NUMBER OF NODES IS 8

STEREO ATTRIBUTES: NONE

L4 SCR 1839

L6 SCR 2040 OR 1929

L8 7111 SEA FILE=REGISTRY SSS FUL L3 AND L4 NOT L6

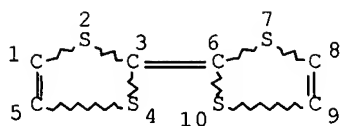
100.0% PROCESSED 840597 ITERATIONS

7111 ANSWERS

SEARCH TIME: 00.00.03

=> d que stat 110

L10 STR

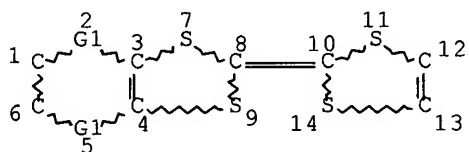


NODE ATTRIBUTES:  
 DEFAULT MLEVEL IS ATOM  
 DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:  
 RING(S) ARE ISOLATED OR EMBEDDED  
 NUMBER OF NODES IS 10

STEREO ATTRIBUTES: NONE

=> d que stat l13  
 L13 STR

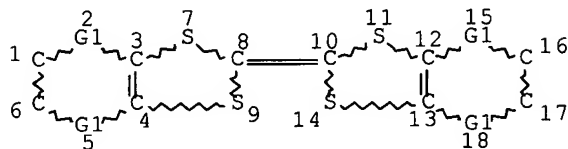


VAR G1=S/O/SE/TE  
 NODE ATTRIBUTES:  
 DEFAULT MLEVEL IS ATOM  
 DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:  
 RING(S) ARE ISOLATED OR EMBEDDED  
 NUMBER OF NODES IS 14

STEREO ATTRIBUTES: NONE

=> d que stat l16  
 L16 STR



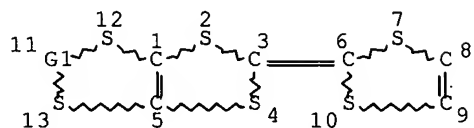
VAR G1=S/O/SE/TE/CH2  
 NODE ATTRIBUTES:  
 DEFAULT MLEVEL IS ATOM  
 DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:  
 RING(S) ARE ISOLATED OR EMBEDDED  
 NUMBER OF NODES IS 18

STEREO ATTRIBUTES: NONE

=> d que stat l19

L19 STR



REP G1=(1-5) CH2

NODE ATTRIBUTES:

DEFAULT MLEVEL IS ATOM

DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:

RING(S) ARE ISOLATED OR EMBEDDED

NUMBER OF NODES IS 13

STEREO ATTRIBUTES: NONE

=> d his nofile

(FILE 'HOME' ENTERED AT 16:50:56 ON 01 JUN 2007)

FILE 'HCAPLUS' ENTERED AT 16:51:04 ON 01 JUN 2007

L1 1 SEA ABB=ON PLU=ON US2004045818/PN  
D SCA  
D IALL  
SEL RN

FILE 'REGISTRY' ENTERED AT 16:51:29 ON 01 JUN 2007

L2 21 SEA ABB=ON PLU=ON (118148-32-6/BI OR 128346-62-3/BI OR  
157289-25-3/BI OR 157289-26-4/BI OR 174421-80-8/BI OR  
214604-40-7/BI OR 25067-58-7/BI OR 31366-25-3/BI OR  
35079-58-4/BI OR 39302-37-9/BI OR 50708-37-7/BI OR  
57512-85-3/BI OR 62921-51-1/BI OR 668421-55-4/BI OR  
668421-56-5/BI OR 668421-57-6/BI OR 668421-58-7/BI OR  
668421-59-8/BI OR 66946-48-3/BI OR 7439-93-2/BI OR  
99159-48-5/BI)  
D SCA

FILE 'LREGISTRY' ENTERED AT 17:09:38 ON 01 JUN 2007

L3 STR  
L4 SCR 1839

FILE 'REGISTRY' ENTERED AT 17:16:11 ON 01 JUN 2007

L5 26 SEA SSS SAM L3 AND L4  
L6 SCR 2040 OR 1929  
L7 20 SEA SSS SAM L3 AND L4 NOT L6  
L8 7111 SEA SSS FUL L3 AND L4 NOT L6  
SAV L8 WEI271/A  
L9 16 SEA ABB=ON PLU=ON L2 AND L8

FILE 'LREGISTRY' ENTERED AT 17:35:54 ON 01 JUN 2007

L10 STR

FILE 'REGISTRY' ENTERED AT 17:41:26 ON 01 JUN 2007

L11 50 SEA SUB=L8 SSS SAM L10  
 L12 5335 SEA SUB=L8 SSS FUL L10

L13 FILE 'LREGISTRY' ENTERED AT 17:46:23 ON 01 JUN 2007  
 STR

L14 FILE 'REGISTRY' ENTERED AT 17:52:36 ON 01 JUN 2007  
 50 SEA SUB=L8 SSS SAM L13  
 L15 1294 SEA SUB=L8 SSS FUL L13  
 SAV L15 WEI271S2/A

L16 FILE 'LREGISTRY' ENTERED AT 17:56:24 ON 01 JUN 2007  
 STR L13

L17 FILE 'REGISTRY' ENTERED AT 17:58:36 ON 01 JUN 2007  
 26 SEA SUB=L8 SSS SAM L16  
 L18 524 SEA SUB=L8 SSS FUL L16  
 SAV L18 WEI271S3/A

L19 FILE 'LREGISTRY' ENTERED AT 18:01:46 ON 01 JUN 2007  
 STR

L20 FILE 'REGISTRY' ENTERED AT 18:04:11 ON 01 JUN 2007  
 49 SEA SUB=L8 SSS SAM L19  
 L21 953 SEA SUB=L8 SSS FUL L19  
 SAV L21 WEI271S4/A  
 L22 1 SEA ABB=ON PLU=ON L2 AND C18H16O4S4/MF  
 D SCA

L23 FILE 'HCAPLUS' ENTERED AT 18:10:26 ON 01 JUN 2007  
 924 SEA ABB=ON PLU=ON L8 (L) DEV+ALL/RL

L24 QUE ABB=ON PLU=ON ELECTROCHEMIC? (2A) DEVICE  
 L25 4 SEA ABB=ON PLU=ON L23 AND L24  
 L26 5761 SEA ABB=ON PLU=ON L8  
 L27 8 SEA ABB=ON PLU=ON L26 AND L24  
 L28 QUE ABB=ON PLU=ON CATHODE? OR ANODE? OR ELECTRODE?  
 L29 507 SEA ABB=ON PLU=ON L26 AND L28  
 L30 188 SEA ABB=ON PLU=ON L8 (L) L28  
 L31 QUE ABB=ON PLU=ON DEVICE  
 L32 13 SEA ABB=ON PLU=ON L30 AND L31  
 L33 18 SEA ABB=ON PLU=ON L25 OR L27 OR L32  
 L34 5014 SEA ABB=ON PLU=ON L12  
 L35 109 SEA ABB=ON PLU=ON L12 (L) L31  
 L36 15145 SEA ABB=ON PLU=ON POLYACETYLENE  
 L37 454 SEA ABB=ON PLU=ON POLYACETYLENE (L) L31  
 L38 3 SEA ABB=ON PLU=ON L35 AND L37  
 L39 22 SEA ABB=ON PLU=ON L34 AND L36  
 L40 5 SEA ABB=ON PLU=ON L39 AND L28 (Formula (2) - searcher's note)  
 L41 1578 SEA ABB=ON PLU=ON L15  
 L42 32 SEA ABB=ON PLU=ON L15 (L) L31  
 L43 15 SEA ABB=ON PLU=ON L42 AND L28 (Formula (3) - searcher's note)  
 L44 1300 SEA ABB=ON PLU=ON L18  
 L45 30 SEA ABB=ON PLU=ON L18 (L) L31  
 L46 15 SEA ABB=ON PLU=ON L45 AND L28 (Formula (4) - searcher's note)  
 L47 1377 SEA ABB=ON PLU=ON L21  
 L48 29 SEA ABB=ON PLU=ON L21 (L) L31  
 L49 15 SEA ABB=ON PLU=ON L48 AND L28 (Formula (5) - searcher's note)  
 L50 2 SEA ABB=ON PLU=ON L22 (Formula (6) - searcher's note)  
 L51 15 SEA ABB=ON PLU=ON L33 AND L28 (Formula (1) - searcher's note)

L52 31 SEA ABB=ON PLU=ON L51 OR L40 OR L43 OR L46 OR L49  
 L53 30 SEA ABB=ON PLU=ON L52 NOT L50  
 L54 4 SEA ABB=ON PLU=ON L53 AND L36  
 L55 30 SEA ABB=ON PLU=ON L53 OR L54

=> fil hcap

FILE 'HCAPLUS' ENTERED AT 18:55:28 ON 01 JUN 2007

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FILE COVERS 1907 - 1 Jun 2007 VOL 146 ISS 24

FILE LAST UPDATED: 31 May 2007 (20070531/ED)

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This file contains CAS Registry Numbers for easy and accurate substance identification.

=> d l50 ibib abs hitstr hitind 1-2

L50 ANSWER 1 OF 2 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2005:1049967 HCAPLUS Full-text

DOCUMENT NUMBER: 143:349949

TITLE: Power system and its manage method

INVENTOR(S): Kuranuki, Masaaki; Inatomi, Yuu

PATENT ASSIGNEE(S): Matsushita Electric Industrial Co., Ltd., Japan

SOURCE: PCT Int. Appl., 32 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent

LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
-----	----	-----	-----	
WO 2005091424	A1	20050929	WO 2005-JP4442	20050314

W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW

RW: BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW,  
 AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ,  
 DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, MC,  
 NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA,  
 GN, GQ, GW, ML, MR, NE, SN, TD, TG

JP 3827709 B2 20060927 JP 2005-518819

200503  
 14

CN 1934745 A 20070321 CN 2005-80008706

200503  
 14

PRIORITY APPLN. INFO.:

JP 2004-78891

A

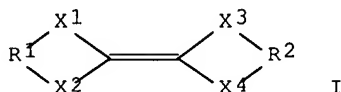
200403  
 18

WO 2005-JP4442

W

200503  
 14

OTHER SOURCE(S): MARPAT 143:349949  
 GI



AB The power system has an electrochem. element, a load, a power generating means, and a charge/discharge control means for the electrochem. element; where the electrochem. element is a secondary battery having a cathode, an anode, and an electrolyte solution or a solid electrolyte and has  $\geq 1$  voltage step on its charge/discharge curve. A threshold voltage is set near the inflection point on 1 of the steps, and the control means controls the charge and discharge of the battery to bring the battery voltage to the threshold voltage. Preferably, the cathode or the anode is I, where R1 and R2 = linear or cyclic aliphatic groups which may contain O, N, S, Si, P, or B atoms, and X1-4 = S, O, to Te; and the power system is for automobiles.

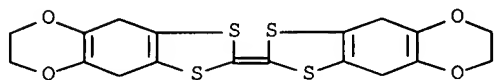
IT 668421-55-4

RL: DEV (Device component use); USES (Uses)

(electrodes for secondary batteries in power systems containing charge/discharge means for automobiles)

RN 668421-55-4 HCAPLUS

CN 1,3-Dithiolo[4,5-g][1,4]benzodioxin, 4,6,7,9-tetrahydro-2-(4,6,7,9-tetrahydro-1,3-dithiolo[4,5-g][1,4]benzodioxin-2-ylidene)- (9CI)  
 (CA INDEX NAME)



IC ICM H01M010-44

ICS G01R031-36; H01M004-60; H02J007-34

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)

IT 668421-55-4

RL: DEV (Device component use); USES (Uses)

(electrodes for secondary batteries in power systems containing charge/discharge means for automobiles)

REFERENCE COUNT: 14 THERE ARE 14 CITED REFERENCES AVAILABLE  
FOR THIS RECORD. ALL CITATIONS AVAILABLE  
IN THE RE FORMAT

L50 ANSWER 2 OF 2 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2004:203235 HCAPLUS Full-text

DOCUMENT NUMBER: 140:238479

TITLE: Electrochemical device

INVENTOR(S): Inatomi, Yuu; Shimada, Mikinari; Hojo, Nobuhiko

PATENT ASSIGNEE(S): Matsushita Electric Industrial Co., Ltd., Japan

SOURCE: U.S. Pat. Appl. Publ., 16 pp.

CODEN: USXXCO

DOCUMENT TYPE: Patent

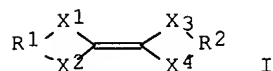
LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. -----	KIND ----	DATE -----	APPLICATION NO. -----	DATE
US 2004045818	A1	20040311	US 2003-648271	200308 27
JP 2004111374	A	20040408	JP 2003-290160	200308 08
EP 1416553	A1	20040506	EP 2003-19484	200308 28
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, SK				
CN 1495939	A	20040512	CN 2003-160286	200308 29
PRIORITY APPLN. INFO.:			JP 2002-250416	A 200208 29

GI



AB The invention concerns an electrochem. device for providing elec. energy by converting an electron transfer involved in an oxidation-reduction reaction into elec. energy comprising a pos. electrode, a neg. electrode and an electrolyte, wherein at least one of the pos. and neg. electrodes comprises a compound having a structure represented by the general formula (I), where R1 and R2 are independent of each other and each represents a linear or cyclic aliphatic group; X1, X2, X3, and X4 are independent of each other and each represents a S atom, an O atom, a Se

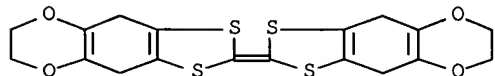
atom, or a Te atom; and the aliphatic group can comprise  $\geq 1$  selected from the group consisting of an O atom, a N atom, a S atom, a Si atom, a P atom, and a B atom.

IT 668421-55-4

RL: DEV (Device component use); USES (Uses)  
(electrochem. device)

RN 668421-55-4 HCAPLUS

CN 1,3-Dithiolo[4,5-g][1,4]benzodioxin, 4,6,7,9-tetrahydro-2-(4,6,7,9-tetrahydro-1,3-dithiolo[4,5-g][1,4]benzodioxin-2-ylidene)- (9CI)  
(CA INDEX NAME)



IC ICM C25B011-04

INCL 204291000

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)  
Section cross-reference(s): 72

IT 7439-93-2, Lithium, uses 25067-58-7D, Polyacetylene, tetrathiafulvalene functionalized 31366-25-3, Tetrathiafulvalene 35079-58-4 39302-37-9, Lithium titanium oxide 50708-37-7, Tetramethyl tetrathiafulvalene 57512-85-3 62921-51-1D, reaction products with polyacetylene 66946-48-3 99159-48-5 118148-32-6 128346-62-3 157289-25-3 157289-26-4 174421-80-8, Cobalt lithium nitride Co<sub>0.4</sub>Li<sub>2.6</sub>N 214604-40-7 668421-55-4 668421-56-5 668421-57-6, Lithium titanium oxide (LiTi<sub>5</sub>O<sub>12</sub>) 668421-58-7 668421-59-8

RL: DEV (Device component use); USES (Uses)  
(electrochem. device)

=> d 155 ibib abs hitstr hitind 1-30

L55 ANSWER 1 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2007:463226 HCAPLUS Full-text

DOCUMENT NUMBER: 146:431572

TITLE: Organic thin film transistor and flat panel display device having the same

INVENTOR(S): Park, Jin-Seong; Suh, Min-Chul; Ahn, Taek

PATENT ASSIGNEE(S): Samsung SDI Co., Ltd., S. Korea

SOURCE: U.S. Pat. Appl. Publ., 16pp.

CODEN: USXXCO

DOCUMENT TYPE: Patent

LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

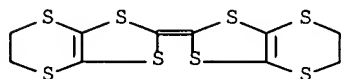
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
-----	----	-----	-----	
US 2007090351	A1	20070426	US 2006-581424	20061017
			KR 2005-99943	A 200510

PRIORITY APPLN. INFO.:



- AB An organic thin film transistor that can control the threshold voltage and reduce leakage current includes: a gate **electrode**; an organic semiconductor layer insulated from the gate **electrode** ; a source **electrode** and a drain **electrode** insulated from the gate **electrode** and elec. connected to the organic semiconductor layer; a gate insulating layer interposed between the gate **electrode** and the organic semiconductor layer; and a hole control layer that is interposed between the gate insulating layer and the organic semiconductor layer. The hole control layer includes a compound having a hole-donor group or a compound having a hole-acceptor group.
- IT **66946-48-3**, Bis(ethylenedithio)tetrathiafulvalene  
 RL: TEM (Technical or engineered material use); USES (Uses)  
 (organic thin film transistor and flat panel display **device**  
 having the same)
- RN 66946-48-3 HCAPLUS
- CN 1,3-Dithiolo[4,5-b][1,4]dithiin, 2-(5,6-dihydro-1,3-dithiolo[4,5-b][1,4]dithiin-2-ylidene)-5,6-dihydro- (CA INDEX NAME)



INCL 257040000; 257066000

- CC 74-13 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)
- IT 84-11-7, Phenanthrenequinone 84-65-1, Anthraquinone 91-19-0, Quinoxaline 92-82-0, Phenazine 92-93-3, 4-Nitrobiphenyl 97-02-9, 2,4-Dinitroaniline 100-01-6, 4-Nitroaniline, uses 117-08-8, Tetrachlorophthalic anhydride 128-69-8, 3,4,9,10-Perylenetetracarboxylic dianhydride 129-79-3, 2,4,7-Trinitrofluorenone 130-15-4, 1,4-Naphthalenedione 275-51-4, Azulene 527-21-9, Tetrafluoro-1,4-benzoquinone 605-71-0, 1,5-Dinitronaphthalene 623-26-7, 1,4-Dicyanobenzene 632-51-9, Tetraphenylethylene 712-74-3, 1,2,4,5-Tetracyanobenzene 961-68-2, 2,4-Dinitrodiphenylamine 1217-45-4, 9,10-Dicyanoanthracene 1953-99-7, Tetrachlorophthalonitrile 2085-33-8, Tris-8-hydroxyquinolinealuminum 4110-35-4, 3,5-Dinitrobenzonitrile 4584-57-0, 4-Dimethylamino-4'-nitrostilbene 15570-45-3, 1,2,3,4-Tetraphenyl-1,3-cyclopentadiene 17420-30-3, 5-Nitroanthranilonitrile 25983-14-6, 2,3,6,7-Tetrachloroquinoxaline 27318-90-7, 1,10-Phenanthroline-5,6-dione **66946-48-3**, Bis(ethylenedithio)tetrathiafulvalene 80509-44-0 126213-51-2, Poly(3,4-ethylenedioxythiophene)  
 RL: TEM (Technical or engineered material use); USES (Uses)  
 (organic thin film transistor and flat panel display **device**  
 having the same)

L55 ANSWER 2 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2007:227234 HCAPLUS Full-text

DOCUMENT NUMBER: 146:299218

TITLE: **Electrode** for use in oxygen reduction

INVENTOR(S): Sotomura, Tadashi; Hashimoto, Mitsuru; Yamada, Yuka

PATENT ASSIGNEE(S): Matsushita Electric Industrial Co., Ltd., Japan

SOURCE: PCT Int. Appl., 44pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent  
 LANGUAGE: Japanese  
 FAMILY ACC. NUM. COUNT: 1  
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2007023964	A1	20070301	WO 2006-JP316773	20060825

W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HN, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KN, KP, KR, KZ, LA, LC, LK, LR, LS, LT, LU, LV, LY, MA, MD, MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RS, RU, SC, SD, SE, SG, SK, SL, SM, SV, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, ZA, ZM, ZW  
 RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG, BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM

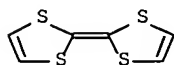
PRIORITY APPLN. INFO.: JP 2005-243846 A 20050825

AB Disclosed is an **electrode** for use in oxygen reduction which can be used as an oxygen **electrode** or air **electrode** in an **electrochem. device** such as an air battery, fuel cell, or electrochem. sensor, can reduce oxygen electrochem. at a noble voltage, and is excellent in stability. The **electrode** comprises a Co tetrapyrizinoporphyrazine derivs. as a catalyst component.

IT 31366-25-3, Tetrathiafulvalene  
 RL: TEM (Technical or engineered material use); USES (Uses)  
 (**electrodes** containing cobalt tetrapyrizinoporphyrazine derivs. for use in oxygen reduction)

RN 31366-25-3 HCAPLUS

CN 1,3-Dithiole, 2-(1,3-dithiol-2-ylidene)- (CA INDEX NAME)



CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)  
 ST fuel cell battery **cathode** catalyst oxygen redn cobalt tetrapyrizinoporphyrazine

IT Nanotubes  
 (carbon; **electrodes** containing cobalt tetrapyrizinoporphyrazine derivs. for use in oxygen reduction)

IT Battery **cathodes**  
 Fuel cell **cathodes**  
 (**electrodes** containing cobalt tetrapyrizinoporphyrazine derivs. for use in oxygen reduction)

IT Carbon black, uses  
 RL: TEM (Technical or engineered material use); USES (Uses)  
 (**electrodes** containing cobalt tetrapyrizinoporphyrazine derivs. for use in oxygen reduction)

IT 11129-60-5, Manganese oxide 12710-12-2, Manganese hydroxide oxide

108916-22-9, Lanthanum manganese strontium oxide (La<sub>0.8</sub>MnSr<sub>0.2</sub>O<sub>3</sub>)  
928144-72-3

RL: CAT (Catalyst use); USES (Uses)

(**electrodes** containing cobalt tetrapyrizinoporphyrazine  
derivs. for use in oxygen reduction)

IT 7440-57-5, Gold, uses 7782-42-5, Graphite, uses 9001-37-0,  
Glucose oxidase 12611-75-5, Nickel steel, uses 31366-25-3  
, Tetrathiafulvalene

RL: TEM (Technical or engineered material use); USES (Uses)

(**electrodes** containing cobalt tetrapyrizinoporphyrazine  
derivs. for use in oxygen reduction)

IT 7440-44-0, Carbon, uses

RL: TEM (Technical or engineered material use); USES (Uses)

(nanotubes; **electrodes** containing cobalt  
tetrapyrizinoporphyrazine derivs. for use in oxygen reduction)

REFERENCE COUNT: 6 THERE ARE 6 CITED REFERENCES AVAILABLE FOR  
THIS RECORD. ALL CITATIONS AVAILABLE IN  
THE RE FORMAT

L55 ANSWER 3 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2006:1352354 HCAPLUS Full-text

DOCUMENT NUMBER: 146:103951

TITLE: Power management system capable of keeping  
residual capacity of **electrochemical**  
**devices** within given range, and  
management method thereof

INVENTOR(S): Kuranuki, Masaaki; Inatomi, Yu

PATENT ASSIGNEE(S): Matsushita Electric Industrial Co., Ltd., Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 21pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent

LANGUAGE: Japanese

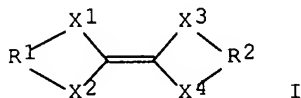
FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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JP 2006351418	A	20061228	JP 2005-177824	200506 17
PRIORITY APPLN. INFO.:				200506 17

OTHER SOURCE(S): MARPAT 146:103951

GI



AB The title system comprises an **electrochem. device** (ECD), e.g., secondary  
batteries, load, and power generation unit, the ECD including a **cathode**, **anode** and  
liquid or solid electrolyte. Preferably, the **cathode** and/or **anode** contain cyclic

comps. I (R1, R2 = linear or cyclic aliphatic group; X1-X4 = S, O, Te) as active mass. The charge/discharge curve of the ECD has a plurality of steps, and a first voltage threshold (VT1) is set at or near an inflection point in an optional first step and second voltage threshold (VT2) at or near an inflection point in a second step on the lower voltage side. The system also comprises (a) means (M1) for judging that ECD voltage is near the VT1 or VT2, (b) a means (M2) for detecting current flowing into or out of the ECD, (c) a means (M3) for integrating outputs of the M2 after the M1 judges that ECD voltage is near the VT1 or VT2 to find a gradient of voltage to the integrated charge/discharge current. It also comprises (d) a charging/discharging controller for judging, when the gradient found by the M3 exceeds a given level, that residual capacity of the ECD increases to the level corresponding to the first step to start discharging electricity from the ECD, or that residual capacity of the ECD decreases to the level corresponding to the second step to start charging the ECD. The system can grasp residual capacity of the ECD relatively easily to keep the residual capacity within a given range. The title method is for managing the above system, comprising steps for implementing the functions (a) to (d).

IT 31366-25-3D, derivs.

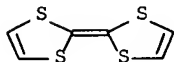
RL: TEM (Technical or engineered material use); **USES (Uses)**

(**cathode** and/or **anode** active mass; power

management system with secondary batteries, load, and power generation unit)

RN 31366-25-3 HCAPLUS

CN 1,3-Dithiole, 2-(1,3-dithiol-2-ylidene)- (CA INDEX NAME)



CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)

IT Battery **electrodes**

Process control

Secondary batteries

(power management system with secondary batteries, load, and power generation unit)

IT 31366-25-3D, derivs.

RL: TEM (Technical or engineered material use); **USES (Uses)**

(**cathode** and/or **anode** active mass; power

management system with secondary batteries, load, and power generation unit)

L55 ANSWER 4 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2006:1153680 HCAPLUS Full-text

DOCUMENT NUMBER: 145:481703

TITLE: Design and operation of a resistance switching memory cell with diode

INVENTOR(S): Krieger, Juri H.; Spitzer, Stuart

PATENT ASSIGNEE(S): Advanced Micro Devices, Inc., USA

SOURCE: U.S. Pat. Appl. Publ., 24pp.

CODEN: USXXCO

DOCUMENT TYPE: Patent

LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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US 2006245235

A1

20061102

US 2005-119973

200505

02

WO 2006118800

A1

20061109

WO 2006-US14797

200604

19

W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA,  
CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI,  
GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM,  
KN, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, LY, MA, MD, MG,  
MK, MN, MW, MX, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT,  
RO, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT,  
TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW

RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU,  
IE, IS, IT, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR,  
BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD,  
TG, BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM,  
ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM

PRIORITY APPLN. INFO.:

US 2005-119973

A

200505

02

AB The invention relates generally to the design and operation of resistance switching memory cells, and in particular to a memory cell with a diode component. Systems and methodologies are provided for forming a diode component operative (e.g., connected in series) with active and passive layer of a resistance switching memory cell to facilitate programming arrays of memory cells created therefrom. Such a diode component can be part of a memory cell having a passive and active layer. Such an arrangement reduces a number of transistor-type voltage controls and associated power consumption, while enabling individual memory cell programming as part of the array. Also, the system provides for an efficient placement of memory cells on a wafer surface, and increases an amount of die space available for circuit design.

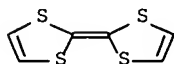
IT 31366-25-3

RL: DEV (Device component use); USES (Uses)

(device active layer; design and operation of a resistance  
switching memory cell with diode)

RN 31366-25-3 HCAPLUS

CN 1,3-Dithiole, 2-(1,3-dithiol-2-ylidene)- (CA INDEX NAME)



INCL 365115000

CC 76-3 (Electric Phenomena)

IT Metallocenes

**Polyacetylenes**, uses

RL: DEV (Device component use); USES (Uses)

(design and operation of a resistance switching memory cell with  
diode)

IT 86-28-2, N-Ethylcarbazole 193-44-2, Tetrathiotetracene 574-93-6,  
Phthalocyanine 670-54-2, Tetracyanoethylene, uses 1518-16-7,  
Tetracyanoquinodimethane 9003-53-6, Polystyrene 9033-83-4,  
Poly(phenylene) 10043-11-5, Boron nitride, uses 12162-21-9,  
Hafnium selenide (HfSe<sub>2</sub>) 12299-51-3, Vanadium selenide (VSe<sub>2</sub>)  
12680-08-9, Lithium titanium sulfide 25013-01-8, Polypyridine  
25038-69-1, Polyphenylacetylene 25067-54-3, Polyfuran

25067-58-7, **Polyacetylene** 25989-14-4,  
 Polydiphenylacetylene 26009-24-5, Poly(p-phenylene vinylene)  
 27290-25-1, Polyphthalocyanine 30604-81-0, Polypyrrole  
**31366-25-3** 82451-55-6, Polyindole 82451-56-7,  
 Polyazulene 108167-10-8 117446-19-2,  
 Hexadecafluorophthalocyanine 126213-51-2,  
 Poly(ethylenedioxythiophene)

RL: DEV (Device component use); USES (Uses)  
 (device active layer; design and operation of a resistance  
 switching memory cell with diode)

IT 7429-90-5, Aluminum, uses 7440-32-6, Titanium, uses

RL: DEV (Device component use); USES (Uses)  
 (**electrode**; design and operation of a resistance  
 switching memory cell with diode)

L55 ANSWER 5 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN  
 ACCESSION NUMBER: 2006:916192 HCAPLUS Full-text  
 DOCUMENT NUMBER: 145:282680  
 TITLE: Organic semiconductor **devices**  
 INVENTOR(S): Takahashi, Yukihiro; Hasegawa, Tatsuo; Abe,  
 Yasushi; Tokura, Yoshinori  
 PATENT ASSIGNEE(S): National Institute of Advanced Industrial  
 Science & Technology, Japan  
 SOURCE: Jpn. Kokai Tokkyo Koho, 11pp.  
 CODEN: JKXXAF  
 DOCUMENT TYPE: Patent  
 LANGUAGE: Japanese  
 FAMILY ACC. NUM. COUNT: 1  
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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JP 2006237271	A	20060907	JP 2005-49759	200502 24
				200502 24

PRIORITY APPLN. INFO.: JP 2005-49759

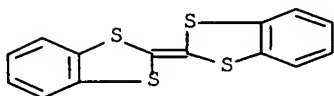
AB The semiconductor **devices** contain organic semiconductor layers and **electrodes**  
 which have elec. conductive charge-transferring complexes made of electron donors  
 and electron acceptors. The **electrodes** have 2 types: those which implant electron  
 in the organic semiconductor layers, and those which implant hole.

IT **24648-13-3**, Dibenzotetrathiafulvalene **31366-25-3**,  
 Tetrathiafulvalene

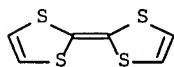
RL: DEV (Device component use); USES (Uses)  
 (TCNQ complex; organic semiconductor **devices** containing organic  
 semiconductor layers and **electrodes** from charge  
 transferring complexes)

RN 24648-13-3 HCAPLUS

CN 1,3-Benzodithiole, 2-(1,3-benzodithiol-2-ylidene)- (CA INDEX NAME)



RN 31366-25-3 HCAPLUS  
 CN 1,3-Dithiole, 2-(1,3-dithiol-2-ylidene)- (CA INDEX NAME)



CC 76-3 (Electric Phenomena)  
 ST org semiconductor **device electrode**; charge  
 transferring complex electron hole implant  
 IT Electron donors  
 (complexes with electron acceptors; organic semiconductor  
**devices** containing organic semiconductor layers and  
**electrodes** from charge transferring complexes)  
 IT Electron acceptors  
 (complexes with electron donors; organic semiconductor  
**devices** containing organic semiconductor layers and  
**electrodes** from charge transferring complexes)  
 IT **Electrodes**  
 (organic semiconductor **devices** containing organic semiconductor  
 layers and **electrodes** from charge transferring  
 complexes)  
 IT Field effect transistors  
 (organic; organic semiconductor **devices** containing organic  
 semiconductor layers and **electrodes** from charge  
 transferring complexes)  
 IT 24648-13-3, Dibenzotetrathiafulvalene 31366-25-3,  
 Tetrathiafulvalene  
 RL: DEV (Device component use); USES (Uses)  
 (TCNQ complex; organic semiconductor **devices** containing organic  
 semiconductor layers and **electrodes** from charge  
 transferring complexes)  
 IT 1518-16-7, TCNQ 29261-33-4, Tetrafluorotetracyanoquinodimethane  
 RL: DEV (Device component use); USES (Uses)  
 (dibenzotetrathiafulvalene complex; organic semiconductor  
**devices** containing organic semiconductor layers and  
**electrodes** from charge transferring complexes)

L55 ANSWER 6 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN  
 ACCESSION NUMBER: 2006:763141 HCAPLUS Full-text  
 DOCUMENT NUMBER: 145:177443  
 TITLE: Organic thin film transistor for flat panel  
 display device  
 INVENTOR(S): Ahn, Taek; Koo, Jae-Bon; Suh, Min-Chul  
 PATENT ASSIGNEE(S): S. Korea  
 SOURCE: U.S. Pat. Appl. Publ., 12 pp.  
 CODEN: USXXCO  
 DOCUMENT TYPE: Patent  
 LANGUAGE: English  
 FAMILY ACC. NUM. COUNT: 1  
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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US 2006169974	A1	20060803	US 2006-338089	200601

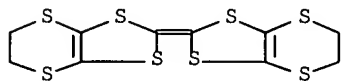
KR 2006087137	A	20060802	KR 2005-7995	24
				200501
				28
CN 1841807	A	20061004	CN 2006-10008937	200601
				28
JP 2006210930	A	20060810	JP 2006-21410	200601
				30
PRIORITY APPLN. INFO.:		KR 2005-7995	A	200501
				28

AB Provided are a thin film transistor, a method of manufacturing the same, and a flat panel display device including the thin film transistor. The thin film transistor includes: a gate **electrode**; source and drain **electrodes** insulated from the gate **electrode**; an organic semiconductor layer that is insulated from the gate **electrode** and elec. connected to the source and drain **electrodes**; an insulating layer that insulates the gate **electrode** from the source and drain **electrodes** or the organic semiconductor layer; and a channel formation-promoting layer that contacts an opposite region of a channel region of the organic semiconductor layer, and contains a compound having a functional group, which fixes elec. charges moving toward the opposite region of the channel region to the opposite region of the channel region. Thus, the thin film transistor has a low threshold voltage and excellent elec. charge mobility.

IT **66946-48-3**, Bis(ethylenedithio)tetrathiafulvalene  
 RL: DEV (Device component use); USES (Uses)  
 (electron donor for organic thin film transistor for flat panel display **device**)

RN 66946-48-3 HCAPLUS

CN 1,3-Dithiolo[4,5-b][1,4]dithiin, 2-(5,6-dihydro-1,3-dithiolo[4,5-b][1,4]dithiin-2-ylidene)-5,6-dihydro- (CA INDEX NAME)



INCL 257040000

CC 74-13 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)  
 Section cross-reference(s): 76

IT 275-51-4, Azulene 632-51-9, Tetraphenylethylene 15570-45-3,  
 1,2,3,4-Tetraphenyl-1,3-cyclopentadiene **66946-48-3**,  
 Bis(ethylenedithio)tetrathiafulvalene 126213-51-2,  
 Poly(3,4-ethylenedioxythiophene)  
 RL: DEV (Device component use); USES (Uses)  
 (electron donor for organic thin film transistor for flat panel display **device**)

L55 ANSWER 7 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2006:734542 HCAPLUS Full-text

DOCUMENT NUMBER: 145:198513

TITLE: Electroluminescent device fabrication by spin coating electroluminescent organometallic complexes on coated substrates

INVENTOR(S): Kathirgamanathan, Poopathy; Ganeshamurugan,



PATENT ASSIGNEE(S): Subramaniam; Price, Richard  
 SOURCE: Oled-T Limited, UK  
 PCT Int. Appl., 51 pp.  
 CODEN: PIXXD2  
 DOCUMENT TYPE: Patent  
 LANGUAGE: English  
 FAMILY ACC. NUM. COUNT: 1  
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2006077402	A1	20060727	WO 2006-GB169	20060119

W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KN, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, LY, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW

RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG, BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM

PRIORITY APPLN. INFO.: GB 2005-1426 A  
 20050122

OTHER SOURCE(S): MARPAT 145:198513

AB Methods of forming electroluminescent devices are described which entail depositing by spin coating a layer of an electroluminescent organometallic complex on a substrate (which is the **anode**) which is coated with a layer of a polymer. The polymer is preferably a conductive or charge-transporting polymer or material.

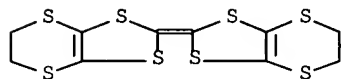
IT 66946-48-3D, derivs.

RL: DEV (Device component use); PEP (Physical, engineering or chemical process); PYP (Physical process); PROC (Process); USES (Uses)

(electroluminescent **device** fabrication by spin coating electroluminescent organometallic complexes on coated substrates)

RN 66946-48-3 HCAPLUS

CN 1,3-Dithiolo[4,5-b][1,4]dithiin, 2-(5,6-dihydro-1,3-dithiolo[4,5-b][1,4]dithiin-2-ylidene)-5,6-dihydro- (CA INDEX NAME)



CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

Section cross-reference(s): 76

IT 86-73-7D, 9H-Fluorene, derivs. 159-66-0D, 9,9'-Spirobi[9H-fluorene], derivs. 193-44-2 905-62-4 1217-45-4, 9,10-Dicyanoanthracene 2085-33-8, Tris(8-hydroxyquinolino)aluminum 4733-39-5, Bathocuproin 5521-31-3D,

derivs. 7429-90-5, Aluminum, uses 7439-93-2, Lithium, uses  
7439-95-4, Magnesium, uses 7440-03-1D, Niobium, compds.  
7440-04-2D, Osmium, compds. 7440-05-3D, Palladium, compds.  
7440-06-4D, Platinum, compds. 7440-16-6D, Rhodium, compds.  
7440-18-8D, Ruthenium, compds. 7440-25-7D, Tantalum, compds.  
7440-32-6D, Titanium, compds. 7440-39-3, Barium, uses  
7440-58-6D, Hafnium, compds. 7440-62-2D, Vanadium, compds.  
7440-70-2, Calcium, uses 7789-24-4, Lithium fluoride, uses  
15082-28-7 17595-05-0 19414-67-6 23467-27-8 25067-59-8,  
Poly(vinylcarbazole) 25135-15-3D, derivs. 25233-30-1,  
Polyaniline 25387-93-3 26009-24-5, Poly(p-phenylenevinylene)-  
31366-25-3D, derivs. 37271-44-6 58280-31-2 58328-31-7, CBP  
58328-31-7D, derivs. 65181-78-4, N,N'-Diphenyl-N,N'-bis(3-  
methylphenyl)-1,1'-biphenyl-4,4'-diamine 66946-48-3D,  
derivs. 95270-88-5D, derivs. 98038-22-3, Aniline-m-sulfanilic  
acid copolymer 121220-44-8, o-Ethylaniline-o-toluidine copolymer  
123847-85-8 124729-98-2 126415-16-5, Aniline-o-anisidine  
copolymer 126415-18-7, o-Aminophenol-aniline copolymer  
126415-20-1, o-Aminophenol-o-toluidine copolymer 126415-22-3,  
o-Phenylenediamine-o-toluidine copolymer 135804-06-7 138372-67-5  
142289-08-5D, derivs. 146162-54-1 148044-16-0 148896-39-3  
150405-69-9 157755-87-8 203642-12-0D, derivs. 214341-85-2D,  
derivs. 221455-80-7 300576-41-4 432042-07-4 432042-08-5  
474974-61-3 474974-62-4 647838-95-7 861532-86-7D,  
[9,9'-Bianthracene]-10,10'-diamine, N-aryl derivs. 863714-50-5  
902119-35-1  
RL: DEV (Device component use); PEP (Physical, engineering or  
chemical process); PYP (Physical process); PROC (Process); USES  
(Uses)

(electroluminescent **device** fabrication by spin coating  
electroluminescent organometallic complexes on coated substrates)

REFERENCE COUNT: 3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR  
THIS RECORD. ALL CITATIONS AVAILABLE IN  
THE RE FORMAT

L55 ANSWER 8 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2006:440184 HCAPLUS Full-text

DOCUMENT NUMBER: 144:479200

TITLE: Organic-complex thin film with bistability for  
nonvolatile memory and electrooptic device  
applications

INVENTOR(S): Yang, Yang; Ouyang, Jianyong; Chu, Chih-Wei

PATENT ASSIGNEE(S): USA

SOURCE: PCT Int. Appl., 25 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent

LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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WO 2006050052	A2	20060511	WO 2005-US38849	200510 27

WO 2006050052 A3 20060629

W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA,  
CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI,  
GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM,

KN, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, LY, MA, MD, MG,  
 MK, MN, MW, MX, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT,  
 RO, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT,  
 TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW  
 RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU,  
 IE, IS, IT, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR,  
 BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD,  
 TG, BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM,  
 ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM

PRIORITY APPLN. INFO.:

US 2004-623721P

P

200410

28

AB The present invention relates to an organic composite material having bistability of an elec. property, electronic or electrooptic devices having the organic composite material and methods of use. An electronic or electrooptic device according to an embodiment of this invention has a 1st **electrode**, a 2nd **electrode** spaced apart from the 1st **electrode**, and an organic composite layer disposed between the 1st **electrode** and the 2nd **electrode**. The organic composite layer is composed of an electron donor material, an electron acceptor material, and a polymer matrix material. The organic composite layer exhibits substantial bistability of an elec. property.

IT 66946-48-3, Bis(ethylenedithio)tetrathiafulvalene

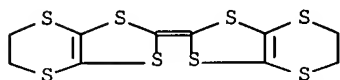
68550-20-9, Bis(methylenedithio)tetrathiafulvalene

RL: DEV (Device component use); TEM (Technical or engineered material use); USES (Uses)

(electron donor; organic-complex thin film with bistability for nonvolatile memory and electrooptic **device** applications)

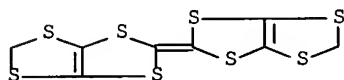
RN 66946-48-3 HCAPLUS

CN 1,3-Dithiolo[4,5-b][1,4]dithiin, 2-(5,6-dihydro-1,3-dithiolo[4,5-b][1,4]dithiin-2-ylidene)-5,6-dihydro- (CA INDEX NAME)



RN 68550-20-9 HCAPLUS

CN [1,3]Dithiolo[4,5-d]-1,3-dithiole, [1,3]dithiolo[4,5-d]-1,3-dithiolylidene- (9CI) (CA INDEX NAME)



IC ICM H01L

CC 76-3 (Electric Phenomena)

Section cross-reference(s): 38, 73

IT 102-54-5, Ferrocene 147-14-8, Copper phthalocyanine 14320-04-8,  
 Zinc(II) phthalocyanine 31366-25-3, Tetrathiafulvalene  
 50708-37-7, Tetramethyltetrathiafulvalene 54489-01-9,  
 Tetraselenafulvalene 55259-49-9, Tetramethyltetraselenafulvalene  
 66946-48-3, Bis(ethylenedithio)tetrathiafulvalene

68550-20-9, Bis(methylenedithio)tetrathiafulvalene  
 101683-17-4, Dimethyl(ethylenedithio)diselenadithiafulvalene  
 152588-53-9, 2,5-Bis(1,3-dithiol-2-ylidene)-1,3,4,6-tetrathiapentalene  
 RL: DEV (Device component use); TEM (Technical or engineered material use); USES (Uses)  
 (electron donor; organic-complex thin film with bistability for nonvolatile memory and electrooptic **device** applications)

L55 ANSWER 9 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN  
 ACCESSION NUMBER: 2006:348925 HCAPLUS Full-text  
 DOCUMENT NUMBER: 145:46342  
 TITLE: Synthesis and characterization of new type molecular wires with tetrathiafulvalene as redox center  
 AUTHOR(S): Wang, Erjing; Li, Hongxiang; Hu, Wenping; Zhu, Daoben  
 CORPORATE SOURCE: Key Laboratory of Organic Solids, Institute of Chemistry, Chinese Academy of Sciences, Beijing, 100080, Peop. Rep. China  
 SOURCE: Journal of Polymer Science, Part A: Polymer Chemistry (2006), 44(8), 2707-2713  
 CODEN: JPACEC; ISSN: 0887-624X  
 PUBLISHER: John Wiley & Sons, Inc.  
 DOCUMENT TYPE: Journal  
 LANGUAGE: English

AB A new type of mol. wire la-c with tetrathiafulvalene (TTF) units was synthesized and characterized. The UV-vis spectra and electrochem. results showed that comparing with PPE, these polymers had smaller HOMO-LUMO band gap, and the HOMO level of polymer la (-5.05 eV) was closer to the work function energy of Au **electrode**. Thermal stability analyses indicated that these polymers had good thermal stability. All of the results showed that the introduction of TTF units made polymers la-c better candidates for mol. wires than PPE.

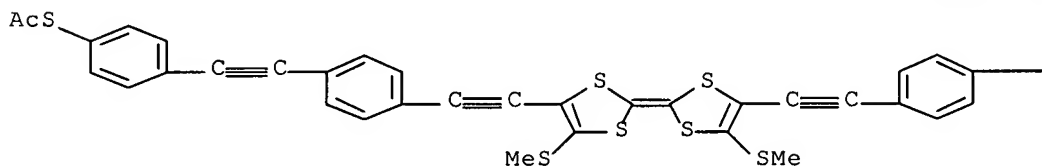
IT 889877-62-7P 889877-63-8P

RL: PRP (Properties); SPN (Synthetic preparation); PREP  
 (Preparation)  
 (model compound; synthesis and characterization of mol. wires with tetrathiafulvalene as redox center)

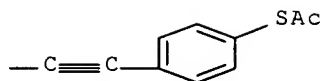
RN 889877-62-7 HCAPLUS

CN Ethanethioic acid, S-[4-[[4-[[2-[4-[[4-[[4-(acetylthio)phenyl]ethynyl]phenyl]ethynyl]-5-(methylthio)-1,3-dithiol-2-ylidene]-5-(methylthio)-1,3-dithiol-4-yl]ethynyl]phenyl]ethynyl]phenyl] ester (9CI) (CA INDEX NAME)

PAGE 1-A

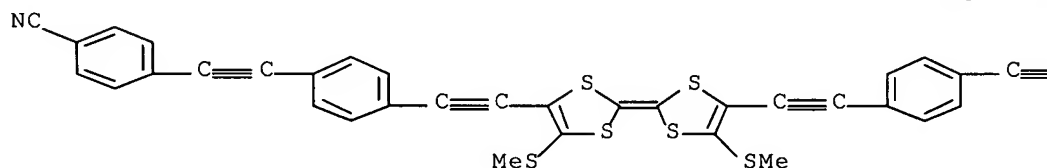


PAGE 1-B

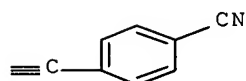


RN 889877-63-8 HCAPLUS  
 CN Benzonitrile, 4-[[4-[[2-[4-[[4-[(4-cyanophenyl)ethynyl]phenyl]ethynyl]-5-(methylthio)-1,3-dithiol-2-ylidene]-5-(methylthio)-1,3-dithiol-4-yl]ethynyl]phenyl]ethynyl]- (9CI) (CA INDEX NAME)

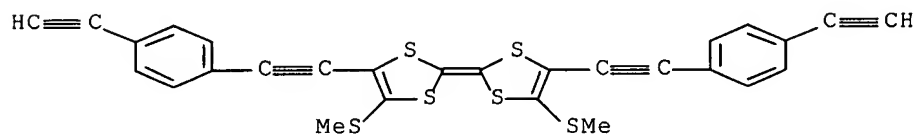
PAGE 1-A



PAGE 1-B



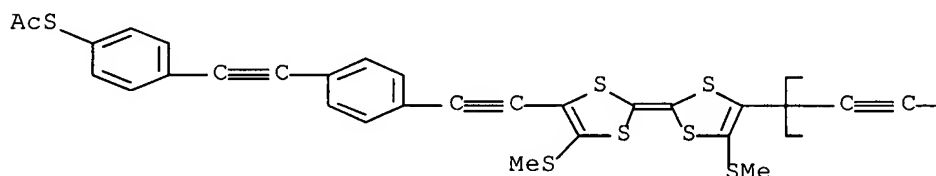
IT 889877-61-6P  
 RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation);  
 RACT (Reactant or reagent)  
 (monomer; synthesis and characterization of mol. wires with  
 tetrathiafulvalene as redox center)  
 RN 889877-61-6 HCAPLUS  
 CN 1,3-Dithiole, 4-[(4-ethynylphenyl)ethynyl]-2-[4-[(4-ethynylphenyl)ethynyl]-5-(methylthio)-1,3-dithiol-2-ylidene]-5-(methylthio)- (9CI) (CA INDEX NAME)



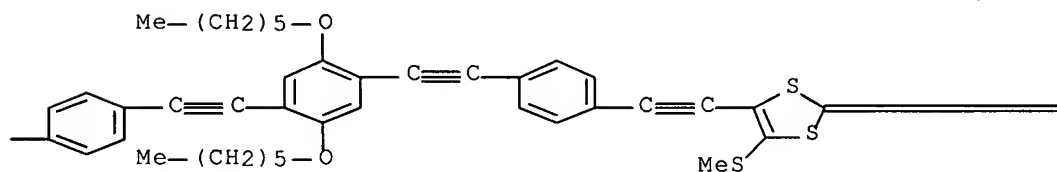
IT 889877-65-0P 889877-66-1P 889877-67-2P  
 RL: PRP (Properties); SPN (Synthetic preparation); PREP  
 (Preparation)  
 (synthesis and characterization of mol. wires with  
 tetrathiafulvalene as redox center)  
 RN 889877-65-0 HCAPLUS  
 CN Poly[[4-(methylthio)-1,3-dithiol-4-yl-2-ylidene][4-(methylthio)-1,3-dithiol-4-yl-2-ylidene]-1,2-ethynediyl-1,4-phenylene-1,2-

ethynediyl[2,5-bis(hexyloxy)-1,4-phenylene]-1,2-ethynediyl-1,4-phenylene-1,2-ethynediyl],  $\alpha$ -[[4-[[4-(acetylthio)phenyl]ethynyl]phenyl]ethynyl]- $\omega$ -[2-[4-[[4-(acetylthio)phenyl]ethynyl]phenyl]ethynyl]-5-(methylthio)-1,3-dithiol-2-ylidene]-5-(methylthio)-1,3-dithiol-4-yl]- (9CI) (CA INDEX NAME)

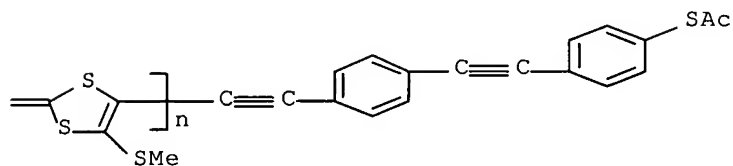
PAGE 1-A



PAGE 1-B



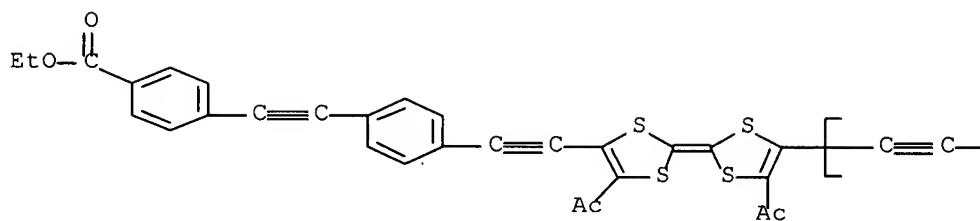
PAGE 1-C



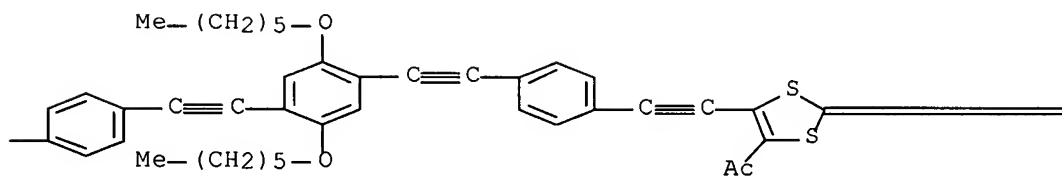
RN 889877-66-1 HCAPLUS

CN Poly[(4-acetyl-1,3-dithiol-4-yl-2-ylidene)(4-acetyl-1,3-dithiol-4-yl-2-ylidene)-1,2-ethynediyl-1,4-phenylene-1,2-ethynediyl[2,5-bis(hexyloxy)-1,4-phenylene]-1,2-ethynediyl-1,4-phenylene-1,2-ethynediyl],  $\alpha$ -[[4-[[4-(ethoxycarbonyl)phenyl]ethynyl]phenyl]ethynyl]- $\omega$ -[5-acetyl-2-[5-acetyl-4-[[4-[[4-(ethoxycarbonyl)phenyl]ethynyl]phenyl]ethynyl]-1,3-dithiol-2-ylidene]-1,3-dithiol-4-yl]- (9CI) (CA INDEX NAME)

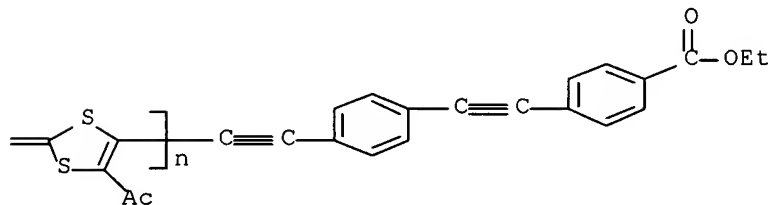
PAGE 1-A



PAGE 1-B



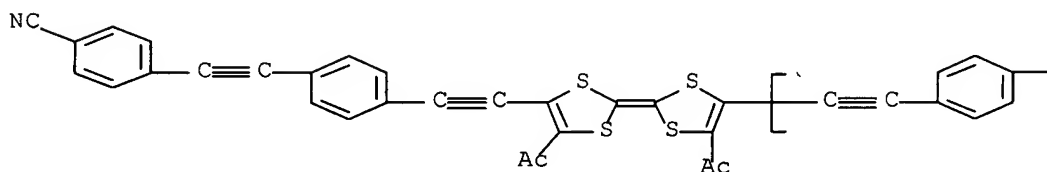
PAGE 1-C



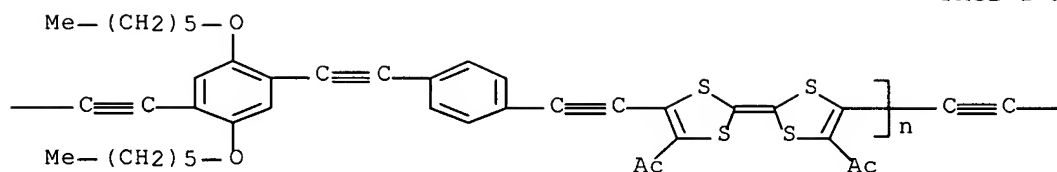
RN 889877-67-2 HCAPLUS

CN Poly[(4-acetyl-1,3-dithiol-4-yl-2-ylidene) (4-acetyl-1,3-dithiol-4-yl-2-ylidene)-1,2-ethynediyl-1,4-phenylene-1,2-ethynediyl[2,5-bis(hexyloxy)-1,4-phenylene]-1,2-ethynediyl-1,4-phenylene-1,2-ethynediyl], α-[[4-[(4-cyanophenyl)ethynyl]phenyl]ethynyl]-ω-[5-acetyl-2-[5-acetyl-4-[[4-[(4-cyanophenyl)ethynyl]phenyl]ethynyl]-1,3-dithiol-2-ylidene]-1,3-dithiol-4-yl]- (9CI) (CA INDEX NAME)

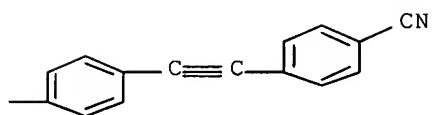
PAGE 1-A



PAGE 1-B



PAGE 1-C

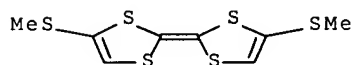


IT 150856-39-6

RL: RCT (Reactant); RACT (Reactant or reagent)  
 (synthesis and characterization of mol. wires with  
 tetrathiafulvalene as redox center)

RN 150856-39-6 HCAPLUS

CN 1,3-Dithiole, 4-(methylthio)-2-[4-(methylthio)-1,3-dithiol-2-ylidene]- (9CI) (CA INDEX NAME)

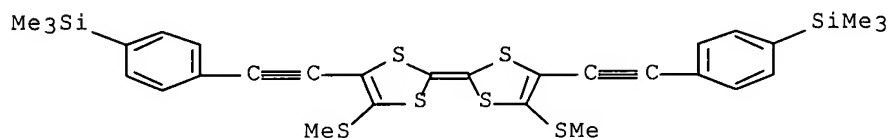


IT 889877-60-5P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation);  
 RACT (Reactant or reagent)  
 (synthesis and characterization of mol. wires with  
 tetrathiafulvalene as redox center)

RN 889877-60-5 HCAPLUS

CN Silane, trimethyl[4-[[5-(methylthio)-2-[4-(methylthio)-5-[[4-(trimethylsilyl)phenyl]ethynyl]-1,3-dithiol-2-ylidene]-1,3-dithiol-4-yl]ethynyl]phenyl]- (9CI) (CA INDEX NAME)



CC 35-7 (Chemistry of Synthetic High Polymers)  
 Section cross-reference(s): 76



ST tetrathiafulvalene **polyacetylene** redox mol wire synthesis  
IT **Polyacetylenes**, preparation  
RL: PRP (Properties); SPN (Synthetic preparation); PREP  
(Preparation)  
(synthesis and characterization of mol. wires with  
tetrathiafulvalene as redox center)  
IT **889877-62-7P 889877-63-8P**  
RL: PRP (Properties); SPN (Synthetic preparation); PREP  
(Preparation)  
(model compound; synthesis and characterization of mol. wires with  
tetrathiafulvalene as redox center)  
IT **889877-61-6P**  
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation);  
RACT (Reactant or reagent)  
(monomer; synthesis and characterization of mol. wires with  
tetrathiafulvalene as redox center)  
IT **889877-64-9DP, acetylthiophenyl-, etoxycarbonylphenyl- or  
cyanophenyl- endcapped 889877-65-0P 889877-66-1P  
889877-67-2P**  
RL: PRP (Properties); SPN (Synthetic preparation); PREP  
(Preparation)  
(synthesis and characterization of mol. wires with  
tetrathiafulvalene as redox center)  
IT **624-73-7, 1,2-Diiodoethane 66228-76-0 150856-39-6**  
RL: RCT (Reactant); RACT (Reactant or reagent)  
(synthesis and characterization of mol. wires with  
tetrathiafulvalene as redox center)  
IT **889877-59-2P 889877-60-5P**  
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation);  
RACT (Reactant or reagent)  
(synthesis and characterization of mol. wires with  
tetrathiafulvalene as redox center)  
REFERENCE COUNT: 45 THERE ARE 45 CITED REFERENCES AVAILABLE  
FOR THIS RECORD. ALL CITATIONS AVAILABLE  
IN THE RE FORMAT

L55 ANSWER 10 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN  
ACCESSION NUMBER: 2006:193499 HCAPLUS Full-text  
DOCUMENT NUMBER: 144:263346  
TITLE: Electron injecting composition including a  
benzoxazole derivative and an electron donating  
organic compound, and light-emitting element and  
light-emitting device using the electron  
injecting composition  
INVENTOR(S): Nakamura, Yasuo; Nomura, Ryoji  
PATENT ASSIGNEE(S): Semiconductor Energy Laboratory Co., Ltd., Japan  
SOURCE: PCT Int. Appl., 52 pp.  
CODEN: PIXXD2  
DOCUMENT TYPE: Patent  
LANGUAGE: English  
FAMILY ACC. NUM. COUNT: 1  
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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WO 2006022194	A1	20060302	WO 2005-JP15110	200508 12

W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA,

CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI,  
 GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM,  
 KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN,  
 MW, MX, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU,  
 SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA,  
 UG, US, UZ, VC, VN, YU, ZA, ZM, ZW  
 RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU,  
 IE, IS, IT, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR,  
 BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD,  
 TG, BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM,  
 ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM  
 JP 2006093673      A      20060406      JP 2005-240682

200508  
 23

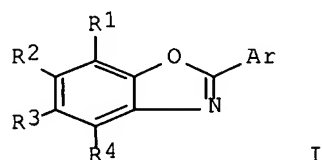
PRIORITY APPLN. INFO.:

JP 2004-242984

A

200408  
 23

OTHER SOURCE(S):      MARPAT 144:263346  
 GI



AB Electron-injecting compns. are described which comprise a benzoxazole derivative indicated by a general formula (I), and an electron donating organic compound, where Ar represents an aryl group, each of R1-4 represents hydrogen, halogen, a cyano group, an alkyl group having 1 to 10 C atoms, a haloalkyl group having 1 to 10 C atoms, an alkoxy group having 1 to 10 C atoms, a substituted or unsubstituted aryl group, or a substituted or unsubstituted heterocyclic group. Light-emitting elements and devices employing the electron-injecting compns. are also discussed.

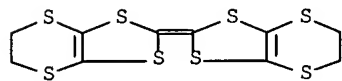
IT **66946-48-3**, Bis(ethylenedithio)tetrathiafulvalene  
**120120-58-3**

RL: DEV (Device component use); TEM (Technical or engineered material use); USES (Uses)

(electron donor; electron injecting composition including benzoxazole derivative and electron donating organic compound, and light-emitting element and light-emitting **device** using electron injecting composition)

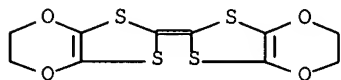
RN 66946-48-3 HCAPLUS

CN 1,3-Dithiolo[4,5-b][1,4]dithiin, 2-(5,6-dihydro-1,3-dithiolo[4,5-b][1,4]dithiin-2-ylidene)-5,6-dihydro- (CA INDEX NAME)



RN 120120-58-3 HCAPLUS

CN 1,3-Dithiolo[4,5-b][1,4]dioxin, 2-(5,6-dihydro-1,3-dithiolo[4,5-b][1,4]dioxin-2-ylidene)-5,6-dihydro- (CA INDEX NAME)



IC ICM H05B033-22  
ICS C09K011-06; H05B033-14  
CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)  
Section cross-reference(s): 28, 76  
IT 7440-32-6, Titanium, uses 25583-20-4, Titanium nitride  
RL: DEV (Device component use); USES (Uses)  
(**anode** layer; electron injecting composition including benzoxazole derivative and electron donating organic compound, and light-emitting element and light-emitting device using electron injecting composition)  
IT 50926-11-9, Indium tin oxide  
RL: DEV (Device component use); USES (Uses)  
(**anode, cathode**; electron injecting composition including benzoxazole derivative and electron donating organic compound, and light-emitting element and light-emitting device using electron injecting composition)  
IT **66946-48-3**, Bis(ethylenedithio)tetrathiafulvalene  
**120120-58-3**  
RL: DEV (Device component use); TEM (Technical or engineered material use); USES (Uses)  
(electron donor; electron injecting composition including benzoxazole derivative and electron donating organic compound, and light-emitting element and light-emitting **device** using electron injecting composition)  
REFERENCE COUNT: 20 THERE ARE 20 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L55 ANSWER 11 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN  
ACCESSION NUMBER: 2006:168071 HCAPLUS Full-text  
DOCUMENT NUMBER: 144:222318  
TITLE: Electronic **device** having an **electrode** with enhanced injection properties  
INVENTOR(S): Brunschweiler, Thomas; Karg, Siegfried F.; Riess, Walter  
PATENT ASSIGNEE(S): International Business Machines Corporation, USA  
SOURCE: U.S. Pat. Appl. Publ., 11 pp.  
CODEN: USXXCO  
DOCUMENT TYPE: Patent  
LANGUAGE: English  
FAMILY ACC. NUM. COUNT: 1  
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
-----	----	-----	-----	
US 2006038170	A1	20060223	US 2005-205232	200508

CN 1738069

A

20060222

CN 2005-10077099

16

200506

15

PRIORITY APPLN. INFO.:

EP 2004-405511

A

200408

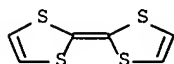
17

AB A method of fabricating an electronic **device** (e.g., electroluminescent **device**) having a first **electrode** is described entailing providing the first **electrode**, depositing a first layer of mol. charge transfer material, which may be an acceptor such as F4-TCNQ, TNF, TeNF, TCNQ, TN9(CN)2F, TCNB, TeCIBQ, TeFTCNQ, DCNQI and TCAQ or a donor such as TTF, TTN, BEDT-TTF, Terpy, Ru(terpy)2 and crystal violet, on the first **electrode**, and crosslinking the mol. charge transfer material (by e.g., UV irradiation). A **device** to fabricate the electronic **device** is also described.

IT 31366-25-3, TTF 66946-48-3, BEDT-TTF  
 RL: DEV (Device component use); USES (Uses)  
 (donor; electronic **device** having crosslinked charge transfer material on **electrode**)

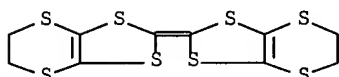
RN 31366-25-3 HCAPLUS

CN 1,3-Dithiole, 2-(1,3-dithiol-2-ylidene)- (CA INDEX NAME)



RN 66946-48-3 HCAPLUS

CN 1,3-Dithiolo[4,5-b][1,4]dithiin, 2-(5,6-dihydro-1,3-dithiolo[4,5-b][1,4]dithiin-2-ylidene)-5,6-dihydro- (CA INDEX NAME)



INCL 257040000; 427058000; 427402000; 427487000; 313504000; 313506000; 428690000; 428917000; 428411100; 118620000

CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)  
 Section cross-reference(s): 76

ST electronic **device** crosslinked charge transfer material

IT Crosslinking  
 (charge transfer material; electronic **device** having crosslinked charge transfer material on **electrode**)

IT Electric apparatus  
 Electroluminescent **devices**  
 Semiconductor **device** fabrication  
 (electronic **device** having crosslinked charge transfer material on **electrode**)

IT 129-79-3, TNF 712-74-3, TCNB 746-53-2 1172-02-7 1518-16-7, TCNQ 15517-55-2 29261-33-4, F4-TCNQ 64374-47-6 70359-39-6 98507-05-2, DCNQI  
 RL: DEV (Device component use); USES (Uses)  
 (acceptor; electronic **device** having crosslinked charge

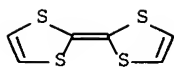
transfer material on **electrode**)  
 IT 548-62-9, Crystal violet 1148-79-4, 2,2':6',2''-Terpyridine  
 31366-25-3, TTF 56348-14-2 66946-48-3, BEDT-TTF  
 143255-97-4  
 RL: DEV (Device component use); USES (Uses)  
 (donor; electronic **device** having crosslinked charge  
 transfer material on **electrode**)

L55 ANSWER 12 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN  
 ACCESSION NUMBER: 2006:143966 HCAPLUS Full-text  
 DOCUMENT NUMBER: 144:224024  
 TITLE: Organic semiconductor **devices** provided  
 with conductive charge transfer complex compound  
**electrodes**  
 INVENTOR(S): Hasegawa, Tatsuo; Takahashi, Yukihiro; Abe,  
 Yasushi; Tokura, Yoshinori  
 PATENT ASSIGNEE(S): National Institute of Advanced Industrial  
 Science & Technology, Japan  
 SOURCE: Jpn. Kokai Tokkyo Koho, 13 pp.  
 CODEN: JKXXAF  
 DOCUMENT TYPE: Patent  
 LANGUAGE: Japanese  
 FAMILY ACC. NUM. COUNT: 1  
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
-----	----	-----	-----	
JP 2006049578	A	20060216	JP 2004-228575	200408 04
PRIORITY APPLN. INFO.: JP 2004-228575				200408 04

AB The title semiconductor **device** has an organic semiconductor layer formed across  
 over 2 **electrodes** which are provided on an insulator layer on a semiconductor  
 substrate, wherein a gate contact is provided directly on the semiconductor  
 substrate. The **electrodes** contain an electron donor in combination to an electron  
 acceptor which has an ionization energy same or similar to that of a semiconductor  
 compound mol. to give a conductive charge transfer complex compound and  
 consequently to give the n-organic HEMTs and semiconductor **devices**. The **devices**  
 may be organic HEMTs, organic electroluminescence diodes, or organic solar cells.

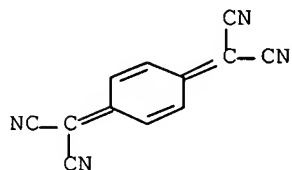
IT 40210-84-2  
 RL: DEV (Device component use); PRP (Properties); USES (Uses)  
 (high electron mobility complex compound, for **electrodes**;  
 organic semiconductor **devices** provided with conductive  
 charge transfer complex compound **electrodes**)  
 RN 40210-84-2 HCAPLUS  
 CN Propanedinitrile, 2,2'-(2,5-cyclohexadiene-1,4-diylidene)bis-,  
 compd. with 2-(1,3-dithiol-2-ylidene)-1,3-dithiole (1:1) (CA INDEX  
 NAME)  
 CM 1  
 CRN 31366-25-3  
 CMF C6 H4 S4



CM 2

CRN 1518-16-7

CMF C12 H4 N4



IT 54928-14-2

RL: PRP (Properties); TEM (Technical or engineered material use);

USES (Uses)

(single crystalline, organic semiconductor; organic semiconductor  
**devices** provided with conductive charge transfer complex  
 compound **electrodes**)

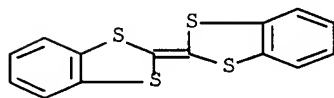
RN 54928-14-2 HCAPLUS

CN Propanedinitrile, 2,2'-(2,5-cyclohexadiene-1,4-diylidene)bis-,  
 compd. with 2-(1,3-benzodithiol-2-ylidene)-1,3-benzodithiolum (1:1)  
 (9CI) (CA INDEX NAME)

CM 1

CRN 24648-13-3

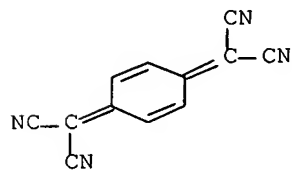
CMF C14 H8 S4



CM 2

CRN 1518-16-7

CMF C12 H4 N4



CC 76-3 (Electric Phenomena)  
 Section cross-reference(s): 38

ST conductive charge transfer complex compd **electrode** org  
 semiconductor **device**; electron donor acceptor ionization  
 energy conductive charge transfer complex

IT Luminescence, electroluminescence  
 (diodes, organic; organic semiconductor **devices** provided with  
 conductive charge transfer complex compound **electrodes**)

IT Electric current-potential relationship  
 (drain current vs. gate or drain voltage; organic semiconductor  
**devices** provided with conductive charge transfer complex  
 compound **electrodes**)

IT Electron acceptors  
 Electron donors  
 (**electrode** composition, for organic semiconductor materials;  
 organic semiconductor **devices** provided with conductive  
 charge transfer complex compound **electrodes**)

IT Charge transfer complexes  
 RL: MOA (Modifier or additive use); PRP (Properties); TEM (Technical  
 or engineered material use); USES (Uses)  
 (**electrode** composition, for organic semiconductor materials;  
 organic semiconductor **devices** provided with conductive  
 charge transfer complex compound **electrodes**)

IT Diodes  
 (organic electroluminescence; organic semiconductor **devices**  
 provided with conductive charge transfer complex compound  
**electrodes**)

IT High-electron-mobility transistors  
 Solar cells  
 (organic semiconductor **devices** provided with conductive  
 charge transfer complex compound **electrodes**)

IT Optical imaging **devices**  
 (organic semiconductor **devices**; organic semiconductor  
**devices** provided with conductive charge transfer complex  
 compound **electrodes**)

IT Semiconductor **devices**  
 Semiconductor materials  
 (organic; organic semiconductor **devices** provided with  
 conductive charge transfer complex compound **electrodes**)

IT 69736-15-8, Polyperylene  
 RL: PRP (Properties)  
 (elec. insulator, for organic semiconductor **devices**; organic  
 semiconductor **devices** provided with conductive charge  
 transfer complex compound **electrodes**)

IT 7440-22-4, Silver, properties  
 RL: PRP (Properties)  
 (gate contact; organic semiconductor **devices** provided with  
 conductive charge transfer complex compound **electrodes**)

IT 40210-84-2  
 RL: DEV (Device component use); PRP (Properties); USES (Uses)  
 (high electron mobility complex compound, for **electrodes**;  
 organic semiconductor **devices** provided with conductive  
 charge transfer complex compound **electrodes**)

IT 54928-14-2  
 RL: PRP (Properties); TEM (Technical or engineered material use);  
 USES (Uses)  
 (single crystalline, organic semiconductor; organic semiconductor  
**devices** provided with conductive charge transfer complex

## compound electrodes)

L55 ANSWER 13 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN  
 ACCESSION NUMBER: 2006:46720 HCAPLUS Full-text  
 DOCUMENT NUMBER: 144:140448  
 TITLE: Two-terminal semiconductor device using abrupt  
 metal-insulator transition semiconductor  
 material  
 INVENTOR(S): Kim, Hyun Tak; Youn, Doo Hyeb; Chae, Byung Gyu;  
 Kang, Kwang Yong; Lim, Yong Sik; Kim, Gyungock;  
 Maeng, Sunglyul; Kim, Seong Hyun  
 PATENT ASSIGNEE(S): Electronics and Telecommunications Research  
 Institute, S. Korea  
 SOURCE: Eur. Pat. Appl., 35 pp.  
 CODEN: EPXXDW  
 DOCUMENT TYPE: Patent  
 LANGUAGE: English  
 FAMILY ACC. NUM. COUNT: 1  
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
EP 1617482	A2	20060118	EP 2004-257769	200412 14
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, PL, SK, BA, HR, IS, YU				
KR 2006006195	A	20060119	KR 2004-55096	200407 15
US 2006011942	A1	20060119	US 2004-11878	200412 13
CN 1722489	A	20060118	CN 2004-10103374	200412 20
JP 2006032898	A	20060202	JP 2004-381971	200412 28
PRIORITY APPLN. INFO.:			KR 2004-55096	A 200407 15

AB Provided is a 2-terminal semiconductor device that uses an abrupt MIT semiconductor material layer. The 2-terminal semiconductor device includes a 1st **electrode** layer, an abrupt MIT semiconductor organic or inorg. material layer having an energy gap <2 eV and holes in a hole level disposed on the 1st **electrode** layer, and a 2nd **electrode** layer disposed on the abrupt MIT semiconductor organic or inorg. material layer. An abrupt MIT is generated in the abrupt MIT semiconductor material layer by a field applied between the 1st **electrode** layer and the 2nd **electrode** layer.

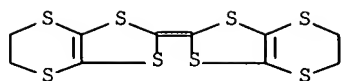
IT 66946-48-3, BEDT-TTF

RL: DEV (Device component use); USES (Uses)  
 (two-terminal semiconductor **device** using abrupt  
 metal-insulator transition semiconductor material)

RN 66946-48-3 HCAPLUS

CN 1,3-Dithiolo[4,5-b][1,4]dithiin, 2-(5,6-dihydro-1,3-dithiolo[4,5-b][1,4]dithiin-2-ylidene)-5,6-dihydro- (CA INDEX NAME)





CC 76-3 (Electric Phenomena)

IT 100-22-1, N,N,N',N'-Tetramethyl-p-phenylenediamine 110-02-1, Thiophene 110-02-1D, Thiophene, derivs. 135-48-8, Pentacene 409-21-2, Silicon carbide (SiC), uses 574-93-6, Phthalocyanine 1303-00-0, Gallium arsenide (GaAs), uses 1303-11-3, Indium arsenide (InAs), uses 1304-82-1, Bismuth telluride (Bi<sub>2</sub>Te<sub>3</sub>) 1306-24-7, Cadmium selenide (CdSe), uses 1306-25-8, Cadmium telluride (CdTe), uses 1313-96-8, Niobium oxide (Nb<sub>2</sub>O<sub>5</sub>) 1314-08-5, Palladium oxide (PdO) 1314-23-4, Zirconium oxide (ZrO<sub>2</sub>), uses 1314-34-7, Vanadium oxide (V<sub>2</sub>O<sub>3</sub>) 1314-35-8, Tungsten oxide (WO<sub>3</sub>), uses 1314-36-9, Yttrium oxide (Y<sub>2</sub>O<sub>3</sub>), uses 1314-61-0, Tantalum oxide (Ta<sub>2</sub>O<sub>5</sub>) 1314-87-0, Lead sulfide (PbS) 1314-91-6, Lead telluride (PbTe) 1314-98-3, Zinc sulfide (ZnS), uses 1315-09-9, Zinc selenide (ZnSe) 1315-11-3, Zinc telluride (ZnTe) 1317-33-5, Molybdenum sulfide (MoS<sub>2</sub>), uses 1317-61-9, Iron oxide (Fe<sub>3</sub>O<sub>4</sub>), uses 1344-28-1, Alumina, uses 1344-54-3, Titanium oxide (TiO<sub>2</sub>) 1518-16-7, Tetracyano-p-quinodimethane 7440-02-0, Nickel, uses 7440-06-4, Platinum, uses 7440-21-3, Silicon, uses 7440-31-5, Tin, uses 7440-36-0, Antimony, uses 7440-42-8, Boron, uses 7440-44-0, Carbon, uses 7440-47-3, Chromium, uses 7440-55-3, Gallium, uses 7440-56-4, Germanium, uses 7440-57-5, Gold, uses 7440-66-6, Zinc, uses 7440-69-9, Bismuth, uses 7631-86-9, Silica, uses 7723-14-0, Phosphorus, uses 7782-49-2, Selenium, uses 10043-11-5, Boron nitride (BN), uses 11092-25-4, Rhodium phosphide (RhP<sub>2</sub>) 11126-12-8, Iron sulfide 11148-21-3 12006-01-8, Ruthenium arsenide phosphide (RuAsP) 12006-29-0, Osmium arsenide (OsAs<sub>2</sub>) 12006-30-3, Ruthenium arsenide (RuAs<sub>2</sub>) 12007-99-7, Calcium boride (CaB<sub>6</sub>) 12010-63-8, Bismuth germanium telluride (Bi<sub>2</sub>GeTe<sub>4</sub>) 12011-54-0, Boron carbide (BC) 12020-69-8, Europium telluride (EuTe) 12022-99-0, Iron silicide (FeSi<sub>2</sub>) 12030-51-2, Iridium sulfide (IrS<sub>2</sub>) 12030-55-6, Iridium selenide (IrSe<sub>2</sub>) 12032-88-1, Manganese telluride (MnTe) 12037-59-1, Osmium phosphide (OsP<sub>2</sub>) 12037-73-9, Ruthenium phosphide (RuP<sub>2</sub>) 12038-20-9, Platinum sulfide (PtS) 12038-21-0D, Platinum sulfide (PtS<sub>2</sub>), platinum-deficient 12038-63-0, Rhenium sulfide (ReS<sub>2</sub>) 12038-64-1, Rhenium selenide (ReSe<sub>2</sub>) 12038-74-3, Rhodium sulfide (RhS<sub>3</sub>) 12038-76-5, Rhodium selenide (RhSe<sub>2</sub>) 12038-77-6, Rhodium selenide (RhSe<sub>3</sub>) 12039-13-3D, Titanium sulfide (TiS<sub>2</sub>), titanium-excess 12039-49-5, Samarium selenide (SmSe) 12039-54-2, Ytterbium selenide (YbSe) 12041-54-2, Aluminum boride (AlB<sub>12</sub>) 12044-16-5, Iron arsenide (FeAs) 12055-23-1, Hafnium oxide (HfO<sub>2</sub>) 12058-20-7D, Molybdenum telluride (MoTe<sub>2</sub>), tellurium-deficient 12063-98-8, Gallium phosphide (GaP), uses 12064-03-8, Gallium antimonide 12067-46-8, Tungsten selenide (WSe<sub>2</sub>) 12068-85-8, Iron sulfide (FeS<sub>2</sub>) 12069-00-0, Lead selenide (PbSe) 12125-57-4, Samarium telluride (SmTe) 12125-58-5, Ytterbium telluride (YbTe) 12162-21-9, Hafnium selenide (HfSe<sub>2</sub>) 12166-20-0, Ruthenium sulfide (RuS<sub>2</sub>) 12166-21-1, Ruthenium selenide (RuSe<sub>2</sub>) 12166-24-4, Ruthenium silicide (Ru<sub>2</sub>Si<sub>3</sub>) 12166-32-4, Zirconium sulfide (Zr<sub>2</sub>S<sub>3</sub>) 12166-47-1D, Zirconium selenide (ZrSe<sub>2</sub>), zirconium-excess 12166-53-9, Zirconium selenide (ZrSe<sub>3</sub>) 12280-05-6, Yttrium boride

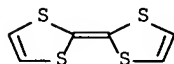
(YB66) 12298-87-2, Manganese silicide (Mn15Si26) 12325-92-7, Manganese silicide (Mn26Si45) 12337-64-3, Rhodium arsenide (RhAs2) 12345-98-1, Europium sulfide (Eu3S4) 12384-18-8, Manganese silicide (Mn11Si19) 12423-80-2D, Titanium sulfide (TiS3), sulfur-deficient 12626-76-5, Iron silicide 13463-67-7, Titanium oxide (TiO2), uses 13494-80-9, Tellurium, uses 15122-76-6, Antimony silver telluride (SbAgTe2) 16150-59-7, Antimony germanium telluride (Sb2GeTe4) 22398-80-7, Indium phosphide (InP), uses 22831-42-1, Aluminum arsenide (AlAs) 25152-52-7, Aluminum antimonide 25617-98-5, Indium nitride (InN) 29678-92-0, Samarium sulfide (SmS) 31366-25-3 34209-23-9, Iridium phosphide (IrP2) 34312-50-0, Technetium sulfide (TcS2) 34312-51-1, Technetium selenide 34312-54-4, Iridium arsenide (IrAs2) 37322-42-2, Samarium boride (SmB6) 52503-00-1, Germanium telluride 54427-07-5, Copper boride 55802-59-0 **66946-48-3**, BEDT-TTF 68898-36-2, Ruthenium phosphide (RuP4) 80146-65-2, Aluminum lithium boride (AlLiB14) 81207-86-5 85906-36-1, Ruthenium arsenide sulfide (RuAsS) 85906-39-4, Osmium phosphide sulfide (OsPS) 85906-40-7, Osmium phosphide selenide (OsPSe) 85906-41-8, Osmium arsenide sulfide (OsAsS) 104934-50-1, Poly(3-hexylthiophene) 104934-51-2, Poly(3-octylthiophene) 104934-53-4, Poly(3-dodecylthiophene) 106070-23-9, Aluminum indium arsenide 106070-25-1, Gallium indium arsenide 109064-29-1, Barium copper yttrium oxide (Ba2Cu3Y07) 113644-78-3 123352-77-2 133087-53-3, Iron manganese silicide ((Fe,Mn)Si2) 133426-01-4, Cobalt iron silicide ((Co,Fe)Si2) 137433-52-4, Potassium fulleride (K6C60) 137751-55-4 140471-84-7, Potassium fulleride (K4C60) 155217-08-6, Cadmium germanium sulfide 263748-31-8  
 RL: DEV (Device component use); USES (Uses)  
 (two-terminal semiconductor **device** using abrupt metal-insulator transition semiconductor material)

L55 ANSWER 14 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN  
 ACCESSION NUMBER: 2005:497080 HCAPLUS Full-text  
 DOCUMENT NUMBER: 143:51861  
 TITLE: Thin film transistor  
 INVENTOR(S): Takenobu, Hiroshi; Iwasa, Yoshihiro  
 PATENT ASSIGNEE(S): Japan Science and Technology Agency, Japan  
 SOURCE: Jpn. Kokai Tokkyo Koho, 10 pp.  
 CODEN: JKXXAF  
 DOCUMENT TYPE: Patent  
 LANGUAGE: Japanese  
 FAMILY ACC. NUM. COUNT: 1  
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2005150410	A	20050609	JP 2003-386114	20031117
PRIORITY APPLN. INFO.:				JP 2003-386114
				20031117

AB A stable thin film transistor having a high mobility comprises a gate **electrode**, a gate insulator film on the gate **electrode**, source and drain **electrodes** on the gate insulator film, and a semiconductor film of C nanotubes and their combination with other material between the source and drain **electrodes**. Specifically, the other material may comprise a fullerene, metal-containing fullerene.

IT 31366-25-3, TTF  
 RL: DEV (Device component use); USES (Uses)  
 (carbon nanotube thin film transistor)  
 RN 31366-25-3 HCAPLUS  
 CN 1,3-Dithiole, 2-(1,3-dithiol-2-ylidene)- (CA INDEX NAME)



IC ICM H01L029-786  
 ICS H01L029-06; H01L051-00  
 CC 76-3 (Electric Phenomena)  
 IT Fullerenes  
 Fullerides  
**Polyacetylenes, uses**  
 RL: DEV (Device component use); USES (Uses)  
 (carbon nanotube thin film transistor)  
 IT 110-02-1D, Thiophene, 3-alkyl, homopolymers 128-65-4 135-48-8,  
 Pentacene 574-93-6, Phthalocyanine 1081-34-1,  
 2,2':5',2''-Terthiophene 1518-16-7, TCNQ 9002-86-2, Polyvinyl  
 chloride 9002-88-4, Polyethylene 9002-98-6, PEI 9003-53-6,  
 Polystyrene 14916-87-1 25067-58-7, **Polyacetylene**  
 25233-34-5, Poly-thiophene 29261-33-4 **31366-25-3, TTF**  
 55259-49-9, TMTSF 66280-99-7, Polythienylenevinylene 78151-58-3  
 88493-55-4,  $\alpha$ -Sexithiophene 97606-53-6 99685-96-8,  
 [5,6]Fullerene-C60-Ih 104934-50-1 105314-21-4 115383-22-7,  
 [5,6]Fullerene-C70-D5h(6) 132814-92-7,  $\alpha$ - $\omega$ -Dihexyl-  
 quaterthiophene 135113-15-4, Fullerene-C76 135113-16-5,  
 Fullerene-C84 136316-32-0, Fullerene-C78 136846-59-8,  
 Fullerene-C82 136846-62-3, Fullerene-C96 137433-42-2  
 146341-33-5 151271-43-1,  $\alpha$ - $\omega$ -Dihexyl-sexithiophene  
 156669-23-7,  $\alpha$ - $\omega$ -Dihexylquinquethiophene 268724-96-5  
 527680-51-9  
 RL: DEV (Device component use); USES (Uses)  
 (carbon nanotube thin film transistor)

L55 ANSWER 15 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN  
 ACCESSION NUMBER: 2005:395664 HCAPLUS Full-text  
 DOCUMENT NUMBER: 142:421857  
 TITLE: Switching device  
 INVENTOR(S): Kawakami, Haruo; Kuroda, Masami; Kato, Hisato;  
 Sekine, Nobuyuki; Yamashiro, Keisuke  
 PATENT ASSIGNEE(S): Fuji Electric Holdings Co., Ltd., Japan  
 SOURCE: PCT Int. Appl., 28 pp.  
 CODEN: PIXXD2  
 DOCUMENT TYPE: Patent  
 LANGUAGE: Japanese  
 FAMILY ACC. NUM. COUNT: 1  
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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WO 2005041318	A1	20050506	WO 2004-JP15519	200410 20

W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA,  
 CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI,  
 GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP,  
 KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW,  
 MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD,  
 SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ,  
 VC, VN, YU, ZA, ZM, ZW  
 RW: BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW,  
 AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ,  
 DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PL,  
 PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ,  
 GW, ML, MR, NE, SN, TD, TG

PRIORITY APPLN. INFO.:

JP 2003-362999

A

200310

23

AB A switching device has high c.d. in ON state as well as high transition voltage. The device exhibits 2 stable resistances to a voltage applied to the **electrodes**. On a substrate, a 1st **electrode** layer, an organic bistable material layer, and a 2nd **electrode** layer are formed as thin films sequentially in order of mention. The organic bistable material layer contains a main component of an electron-transporting organic bistable material and an additive of an electron-donating compound

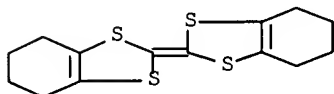
IT 35079-58-4 66946-48-3, BEDT-TTF

RL: DEV (Device component use); USES (Uses)

(electron donor layer; switching **devices** containing organic bistable material layers for organic EL **devices** and high.-d. memory **devices**)

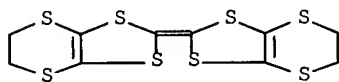
RN 35079-58-4 HCAPLUS

CN 1,3-Benzodithiolo, 4,5,6,7-tetrahydro-2-(4,5,6,7-tetrahydro-1,3-benzodithiol-2-ylidene)- (9CI) (CA INDEX NAME)



RN 66946-48-3 HCAPLUS

CN 1,3-Dithiolo[4,5-b][1,4]dithiin, 2-(5,6-dihydro-1,3-dithiolo[4,5-b][1,4]dithiin-2-ylidene)-5,6-dihydro- (CA INDEX NAME)



IC ICM H01L051-00

CC 76-3 (Electric Phenomena)

IT 7429-90-5, Aluminum, uses 7440-57-5, Gold, uses

RL: DEV (Device component use); USES (Uses)

(**electrode**; switching devices containing organic bistable material layers for organic EL devices and high.-d. memory devices)

IT 35079-58-4 66946-48-3, BEDT-TTF 71938-96-0

RL: DEV (Device component use); USES (Uses)

(electron donor layer; switching **devices** containing organic bistable material layers for organic EL **devices** and high.-d. memory **devices**)

REFERENCE COUNT: 2 THERE ARE 2 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L55 ANSWER 16 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2005:209590 HCAPLUS Full-text

DOCUMENT NUMBER: 142:289758

TITLE: Memory devices based on electric field programmable films

INVENTOR(S): Yang, Yang; Ouyang, Jianyong; Szmanda, Charles R.

PATENT ASSIGNEE(S): The Regents of the University of California, USA; Rohm and Haas Company

SOURCE: Eur. Pat. Appl., 25 pp.

CODEN: EPXXDW

DOCUMENT TYPE: Patent

LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. -----	KIND ----	DATE -----	APPLICATION NO. -----	DATE
EP 1513159	A2	20050309	EP 2004-255350	200409 03
EP 1513159 R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, PL, SK, HR	A3	20050921		
CA 2479317	A1	20050303	CA 2004-2479317	200408 26
KR 2005025088	A	20050311	KR 2004-70529	200409 03
JP 2005101594	A	20050414	JP 2004-256613	200409 03
CN 1651496	A	20050810	CN 2004-10082230	200409 03
PRIORITY APPLN. INFO.:			US 2003-500082P	P 200309 03

AB A composition for the formation of an elec. field programmable film, the composition comprising a matrix precursor composition or a dielec. matrix material, wherein the dielec. matrix material comprises an organic polymer and/or an inorg. oxide; and an electron donor and an electron acceptor of a type and in an amount effective to provide elec. field programming. The films are of utility in data storage devices.

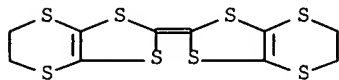
IT **66946-48-3**, Bis(ethylenedithio)tetrathiafulvalene

RL: DEV (Device component use); USES (Uses)

(electron donor; memory **devices** based on elec. field programmable films from dielec. matrix composites)

RN 66946-48-3 HCAPLUS

CN 1,3-Dithiolo[4,5-b][1,4]dithiin, 2-(5,6-dihydro-1,3-dithiolo[4,5-b][1,4]dithiin-2-ylidene)-5,6-dihydro- (CA INDEX NAME)



IC ICM G11C013-02  
 CC 76-14 (Electric Phenomena)  
 Section cross-reference(s): 38, 66  
 IT 7429-90-5, Aluminum, processes 50926-11-9, ITO  
 RL: DEV (Device component use); PEP (Physical, engineering or chemical process); PYP (Physical process); PROC (Process); USES (Uses)  
 (electrode; memory devices based on elec. field programmable films from dielec. matrix composites)  
 IT 86-28-2, N-Ethylcarbazole 87-85-4, Hexamethylbenzene 106-50-3, p-Phenylenediamine, uses 193-44-2, Tetrathiotetracene 31366-25-3, Tetrathiafulvalene 49868-52-2 55259-49-9 56366-76-8 **66946-48-3**, Bis(ethylenedithio)tetrathiafulvalene  
 RL: DEV (Device component use); USES (Uses)  
 (electron donor; memory devices based on elec. field programmable films from dielec. matrix composites)

L55 ANSWER 17 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN  
 ACCESSION NUMBER: 2004:677671 HCAPLUS Full-text  
 DOCUMENT NUMBER: 141:197155  
 TITLE: OLED device with a performance enhancing layer based on chemical reducing materials  
 INVENTOR(S): Liao, Liang-sheng; Madathil, Joseph K.; Klubek, Kevin P.; Comfort, Dustin L.; Tang, Ching W.  
 PATENT ASSIGNEE(S): Eastman Kodak Company, USA  
 SOURCE: Eur. Pat. Appl., 20 pp.  
 CODEN: EPXXDW  
 DOCUMENT TYPE: Patent  
 LANGUAGE: English  
 FAMILY ACC. NUM. COUNT: 1  
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
EP 1447862	A2	20040818	EP 2004-75315	20040202
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, SK				
US 6781149	B1	20040824	US 2003-366835	20030214
KR 2004073986	A	20040821	KR 2004-9481	20040213
JP 2004247309	A	20040902	JP 2004-36976	200402

CN 1610464

A

20050427

CN 2004-10007822

13

200402

14

PRIORITY APPLN. INFO.:

US 2003-366835

A

200302

14

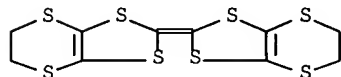
AB Organic light-emitting devices with improved performance are described which comprise an **anode** formed over a substrate; a hole-transporting layer formed over the **anode**; a light-emitting layer formed over the hole-transporting layer for producing light in response to hole-electron recombination; a performance-enhancing layer formed over the light-emitting layer including one or more chemical reducing materials selected to improve the performance of the organic light-emitting device; an electron-transporting layer formed over the performance-enhancing layer; and a **cathode** formed over the electron-transporting layer.

IT **66946-48-3**, Bis(ethylenedithio)tetrathiafulvalene  
**66946-48-3D**, Bis(ethylenedithio)tetrathiafulvalene, derivs.

RL: DEV (Device component use); USES (Uses)  
(performance-enhancing layer; fabrication of OLED **device**  
with performance-enhancing layer based on chemical reducing  
materials)

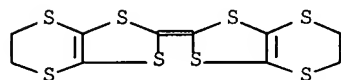
RN 66946-48-3 HCAPLUS

CN 1,3-Dithiolo[4,5-b][1,4]dithiin, 2-(5,6-dihydro-1,3-dithiolo[4,5-b][1,4]dithiin-2-ylidene)-5,6-dihydro- (CA INDEX NAME)



RN 66946-48-3 HCAPLUS

CN 1,3-Dithiolo[4,5-b][1,4]dithiin, 2-(5,6-dihydro-1,3-dithiolo[4,5-b][1,4]dithiin-2-ylidene)-5,6-dihydro- (CA INDEX NAME)



IC ICM H01L051-20

CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

Section cross-reference(s): 76

IT 31366-25-3, Tetrathiafulvalene 31366-25-3D, Tetrathiafulvalene, derivs. **66946-48-3**, Bis(ethylenedithio)tetrathiafulvalene  
**66946-48-3D**, Bis(ethylenedithio)tetrathiafulvalene, derivs.

RL: DEV (Device component use); USES (Uses)  
(performance-enhancing layer; fabrication of OLED **device**  
with performance-enhancing layer based on chemical reducing  
materials)

L55 ANSWER 18 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2004:677670 HCAPLUS Full-text

DOCUMENT NUMBER: 141:181714

TITLE: Forming an OLED device with a performance-enhancing layer based on chemical reducing materials  
 INVENTOR(S): Boroson, Michael L.; Liao, Liang-sheng  
 PATENT ASSIGNEE(S): Eastman Kodak Company, USA  
 SOURCE: Eur. Pat. Appl., 18 pp.  
 CODEN: EPXXDW  
 DOCUMENT TYPE: Patent  
 LANGUAGE: English  
 FAMILY ACC. NUM. COUNT: 1  
 PATENT INFORMATION:

PATENT NO. -----	KIND ----	DATE -----	APPLICATION NO. -----	DATE
EP 1447861	A2	<u>20040818</u>	EP 2004-75314	20040202
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, SK				
US 2004161695	A1	20040819	US 2003-366945	20030214
US 6824950	B2	20041130		
KR 2004073995	A	20040821	KR 2004-9616	20040213
JP 2004247310	A	20040902	JP 2004-36993	20040213
CN 1535083	A	20041006	CN 2004-10005063	20040216
PRIORITY APPLN. INFO.: US 2003-366945				A 20030214

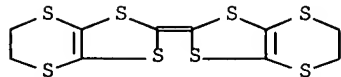
AB Methods for forming an organic light-emitting device with improved performance are discussed which entail forming an **anode** over a substrate; providing a donor element including light-emitting materials and positioning such donor element in a material-transferring relationship with the substrate; illuminating the donor element with radiation to cause the transfer of light-emitting material to deposit the light-emitting material and form a light-emitting layer over the **anode**; forming a performance-enhancing layer over the light-emitting layer including 1 or more chemical reducing materials selected to improve the performance of the organic light-emitting device; forming an electron-transporting layer over the performance-enhancing layer; and forming a **cathode** over the electron-transporting layer. The performance-enhancing layer may include one or more metallic materials selected from alkali metals, alkaline earth metals, and lanthanide metals, or one or more organic chemical reducing materials selected from bis(ethylenedithio)tetrathiafulvalene, tetrathiafulvalene, and their derivs.

IT **66946-48-3**, Bis(ethylenedithio)tetrathiafulvalene  
**66946-48-3D**, Bis(ethylenedithio)tetrathiafulvalene, derivs.  
 RL: DEV (Device component use); USES (Uses)  
 (performance-enhancing layer; fabrication of OLED device with performance-enhancing layer based on chemical reducing materials)

RN 66946-48-3 HCAPLUS  
 CN 1,3-Dithiolo[4,5-b][1,4]dithiin, 2-(5,6-dihydro-1,3-dithiolo[4,5-

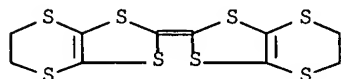


b][1,4]dithiin-2-ylidene)-5,6-dihydro- (CA INDEX NAME)



RN 66946-48-3 HCAPLUS

CN 1,3-Dithiololo[4,5-b][1,4]dithiin, 2-(5,6-dihydro-1,3-dithiololo[4,5-b][1,4]dithiin-2-ylidene)-5,6-dihydro- (CA INDEX NAME)



IC ICM H01L051-20

ICS H01L051-40

CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

Section cross-reference(s): 76

IT 7440-39-3, Barium, uses 31366-25-3, Tetrathiafulvalene

31366-25-3D, Tetrathiafulvalene, derivs. **66946-48-3**,

Bis(ethylenedithio)tetrathiafulvalene **66946-48-3D**,

Bis(ethylenedithio)tetrathiafulvalene, derivs.

RL: DEV (Device component use); USES (Uses)

(performance-enhancing layer; fabrication of OLED **device**

with performance-enhancing layer based on chemical reducing materials)

L55 ANSWER 19 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2004:632970 HCAPLUS Full-text

DOCUMENT NUMBER: 141:164594

TITLE: Organic electroluminescent device driven at low voltage

INVENTOR(S): Suh, Min-Chul

PATENT ASSIGNEE(S): Samsung SDI Co., Ltd., Japan

SOURCE: U.S. Pat. Appl. Publ., 11 pp.

CODEN: USXXCO

DOCUMENT TYPE: Patent

LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
-----	----	-----	-----	
US 2004150330	A1	20040805	US 2004-757471	20040115
KR 2004070514	A	20040811	KR 2003-6617	20030203
CN 1610469	A	20050427	CN 2004-10038754	200402

PRIORITY APPLN. INFO.:

KR 2003-6617

03  
A  
200302  
03

AB An organic electroluminescent device is described comprising a substrate; a first **electrode** to define a pixel region on the substrate; multiple organic film layers to perform light emission on the first **electrode**; and a second **electrode** formed on the multiple organic film layers, wherein the multiple organic film layers comprise an emitting layer; and at least one of a hole injection layer and a hole transfer layer; wherein the at least one of the hole injection layer and the hole transfer layer comprises an electron acceptor material. The organic electroluminescent device may have an improved lifetime and may be driven at a low voltage.

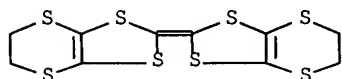
IT **66946-48-3**, Bis(ethylenedithio)tetrathiafulvalene

RL: DEV (Device component use); USES (Uses)

(electron donor; organic electroluminescent **device** having  
luminescence quenching effect driven at low voltage)

RN 66946-48-3 HCAPLUS

CN 1,3-Dithiolo[4,5-b][1,4]dithiin, 2-(5,6-dihydro-1,3-dithiolo[4,5-b][1,4]dithiin-2-ylidene)-5,6-dihydro- (CA INDEX NAME)



IC ICM H05B033-00

INCL 313506000

CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

Section cross-reference(s): 76

IT 275-51-4, Azulene 632-51-9, Tetraphenylethylene 15570-45-3,  
1,2,3,4-Tetraphenyl-1,3-cyclopentadiene **66946-48-3**,  
Bis(ethylenedithio)tetrathiafulvalene 126213-51-2,  
Poly(3,4-ethylene-dioxythiophene)

RL: DEV (Device component use); USES (Uses)

(electron donor; organic electroluminescent **device** having  
luminescence quenching effect driven at low voltage)

L55 ANSWER 20 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2003:173208 HCAPLUS Full-text

DOCUMENT NUMBER: 138:408228

TITLE: High-conductivity organic metals as  
**electrode** materials

AUTHOR(S): Pospelov, Alexander P.; Ved, Marina V.;  
Sakhnenko, Nikolay D.; Alexandrov, Yuriy L.;  
Shtefan, Viktoria V.; Kravchenko, Andrey V.;  
Kamarchuk, Gennadiy V.

CORPORATE SOURCE: National Technical University Kharkov  
Polytechnical Institute, Kharkov, Ukraine

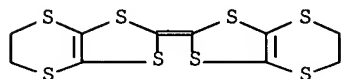
SOURCE: Materials Science (2002), 20(3), 65-72  
CODEN: MSCJDS; ISSN: 0137-1339

PUBLISHER: Wroclaw University of Technology, Centre of  
Advanced Materials and Nanotechnology

DOCUMENT TYPE: Journal

LANGUAGE: English

- AB **Electrode** properties of TCNQ (7,7,8,8- tetracyanoquinodimethane) and BEDT-TTF (bis- (ethylenedithio)tetrathiafulvalene) derivs. are considered. The BEDT-TTF-based organic **electrode** materials were produced by electrochem. technique. **Electrodes** with TCNQ salts were obtained by thermal or evaporation method. Polarization and impedance investigations have shown the high **electrode** activity of the BEDT-TTF based materials in irreversible electrochem. processes. TCNQ-based OM sensitivity to pH as well as **electrode** surface resistance vary depending on gaseous phase composition The latter circumstance is quite prospective for applications of organic metals in anal. control **devices**.
- IT 66946-48-3, Bis-(ethylenedithio)tetrathiafulvalene  
 RL: DEV (Device component use); PRP (Properties); USES (Uses)  
 (derivs.; high-conductivity organic metals as **electrode** materials)
- RN 66946-48-3 HCAPLUS
- CN 1,3-Dithiolo[4,5-b][1,4]dithiin, 2-(5,6-dihydro-1,3-dithiolo[4,5-b][1,4]dithiin-2-ylidene)-5,6-dihydro- (CA INDEX NAME)



- CC 72-2 (Electrochemistry)  
 Section cross-reference(s): 29, 79
- ST molten salt org metal **electrode** material electrosynthesis
- IT Gas sensors  
 (Pt **electrode** modified with evaporated organic metals for gases)
- IT Sensors  
 (electrochem.; Pt **electrode** modified with evaporated organic metals for gases)
- IT **Electrodes**  
 (high-conductivity organic metals as **electrode** materials)
- IT Salts, uses  
 RL: NUU (Other use, unclassified); USES (Uses)  
 (molten; high-conductivity organic metals as **electrode** materials)
- IT Electric capacitance  
 (of Pt **electrode** modified with (ET)2Mo6O19 in H2SO4)
- IT Cyclic voltammetry  
 (of Pt **electrodes** bare and modified with (ET)2Mo6O19 in H2SO4)
- IT 7664-93-9, Sulfuric acid, uses  
 RL: NUU (Other use, unclassified); USES (Uses)  
 (cyclic voltammetry of Pt **electrodes** bare and modified with (ET)2Mo6O19 in H2SO4)
- IT 66946-48-3, Bis-(ethylenedithio)tetrathiafulvalene  
 RL: DEV (Device component use); PRP (Properties); USES (Uses)  
 (derivs.; high-conductivity organic metals as **electrode** materials)
- IT 12390-22-6  
 RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); RCT (Reactant); PROC (Process); RACT (Reactant or reagent)  
 (electrosynthesis of high-conductivity organic metals as **electrode** materials in solution containing)
- IT 68-12-2, DMF, uses 2537-36-2, Tetramethylammonium perchlorate  
 RL: NUU (Other use, unclassified); USES (Uses)  
 (electrosynthesis of high-conductivity organic metals as **electrode** materials in solution containing)

IT 134116-05-5P  
RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); PROC (Process)  
(electrosynthesis of high-conductivity organic metals as **electrode** materials in solution containing tetracyanoquinodimethane or bis-(ethylenedithio)tetrathiafulvalene derivs. on)

IT 7440-06-4, Platinum, uses  
RL: DEV (Device component use); USES (Uses)  
(electrosynthesis of high-conductivity organic metals as **electrode** materials in solution containing tetracyanoquinodimethane or bis-(ethylenedithio)tetrathiafulvalene derivs. on)

IT 1518-16-7  
RL: DEV (Device component use); PRP (Properties); USES (Uses)  
(high-conductivity organic metals as **electrode** materials)

REFERENCE COUNT: 7 THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L55 ANSWER 21 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2002:144492 HCAPLUS Full-text

DOCUMENT NUMBER: 137:127441

TITLE: Deposition of organic **electrodes** based on wet process for organic **devices**

AUTHOR(S): Saito, Kazuhiro; Kobayashi, Shunsuke

CORPORATE SOURCE: National Institute of Advanced Industrial Science and Technology, Tsukuba-shi, Ibaraki, 305-8568, Japan

SOURCE: Applied Physics Letters (2002), 80(8), 1489-1491

CODEN: APPLAB; ISSN: 0003-6951

PUBLISHER: American Institute of Physics

DOCUMENT TYPE: Journal

LANGUAGE: English

AB Patterned organic **electrodes** of charge-transfer complexes were deposited based on a printing method and solution chemical without a vacuum and high temperature. The deposited organic **electrodes** showed large work functions, and they were examined as upper **electrodes** of organic photovoltaic cells. It is found that the charge-transfer complexes can be used as wiring material instead of metals without secondary treatment. In comparison with the cells using the conventional metals, a few different properties were observed for those with organic **electrodes**. The differences are assignable to the difference between the organic-organic and the organic-inorg. contacts.

IT 40210-84-2P, TTF-TCNQ

RL: DEV (Device component use); PNU (Preparation, unclassified); PRP (Properties); PREP (Preparation); USES (Uses)

(deposition of organic **electrodes** based on wet process for organic **devices**)

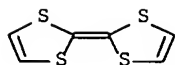
RN 40210-84-2 HCAPLUS

CN Propanedinitrile, 2,2'-(2,5-cyclohexadiene-1,4-diylidene)bis-, compd. with 2-(1,3-dithiol-2-ylidene)-1,3-dithiole (1:1) (CA INDEX NAME)

CM 1

CRN 31366-25-3

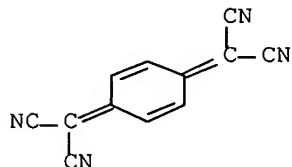
CMF C6 H4 S4



CM 2

CRN 1518-16-7

CMF C12 H4 N4



CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)  
 ST wet deposition org **electrode** charge transfer complex org  
**device**  
 IT Photoelectric **devices**  
 (deposition of organic **electrodes** based on wet process for  
 organic **devices**)  
 IT Charge transfer complexes  
 RL: DEV (Device component use); PNU (Preparation, unclassified); PRP  
 (Properties); PREP (Preparation); USES (Uses)  
 (deposition of organic **electrodes** based on wet process for  
 organic **devices**)  
 IT **Electrodes**  
 (of organic photovoltaic cells; deposition of organic  
**electrodes** based on wet process for organic **devices**  
 )  
 IT **40210-84-2P**, TTF-TCNQ 84632-22-4P  
 RL: DEV (Device component use); PNU (Preparation, unclassified); PRP  
 (Properties); PREP (Preparation); USES (Uses)  
 (deposition of organic **electrodes** based on wet process for  
 organic **devices**)  
 REFERENCE COUNT: 20 THERE ARE 20 CITED REFERENCES AVAILABLE  
 FOR THIS RECORD. ALL CITATIONS AVAILABLE  
 IN THE RE FORMAT

L55 ANSWER 22 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN  
 ACCESSION NUMBER: 1999:147267 HCAPLUS Full-text  
 DOCUMENT NUMBER: 130:189145  
 TITLE: Method of manufacturing organic/polymer  
 electroluminescent device  
 INVENTOR(S): Zyung, Taehyoung; Jung, Sang-don; Choi,  
 Kang-hoon  
 PATENT ASSIGNEE(S): Electronics and Telecommunications Research  
 Institute, S. Korea  
 SOURCE: U.S., 6 pp.  
 CODEN: USXXAM  
 DOCUMENT TYPE: Patent  
 LANGUAGE: English  
 FAMILY ACC. NUM. COUNT: 1

## PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 5876786	A	19990302	US 1997-919929	19970828
PRIORITY APPLN. INFO.:			KR 1996-35936	A 19960828

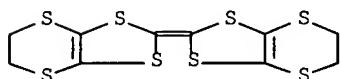
AB Methods of manufacturing electroluminescent devices are described which entail preparing a transparent substrate; depositing a transparent layer on the substrate; forming a plurality of transparent **electrodes** on selected portions of the substrate by patterning the layer; depositing a first film comprising a first charge transfer material on the resulting structure; depositing an emissive layer on the first film; depositing a second film comprising a second charge transfer material on the emissive layer; depositing a metal layer on the second film; and forming a plurality of metal **electrodes** on selected portions of the second film by patterning the metal layer. Forming the film consisting of a charge transfer complex or charge transfer salt between the organic/polymer electroluminescent layer and **electrodes** for injecting electrons and holes increases the electroluminescent quantum efficiency.

IT 66946-48-3, Bis(ethylenedithio)-tetrathiafulvalene  
118148-29-1 120120-58-3

RL: DEV (Device component use); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses)  
(organic/polymer electroluminescent **device** fabrication)

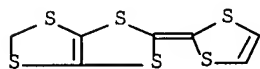
RN 66946-48-3 HCAPLUS

CN 1,3-Dithiolo[4,5-b][1,4]dithiin, 2-(5,6-dihydro-1,3-dithiolo[4,5-b][1,4]dithiin-2-ylidene)-5,6-dihydro- (CA INDEX NAME)



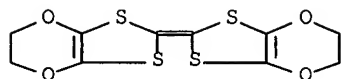
RN 118148-29-1 HCAPLUS

CN [1,3]Dithiolo[4,5-d]-1,3-dithiole, 1,3-dithiol-2-ylidene- (9CI) (CA INDEX NAME)



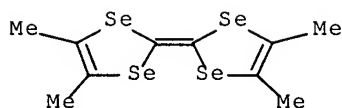
RN 120120-58-3 HCAPLUS

CN 1,3-Dithiolo[4,5-b][1,4]dioxin, 2-(5,6-dihydro-1,3-dithiolo[4,5-b][1,4]dioxin-2-ylidene)-5,6-dihydro- (CA INDEX NAME)



IC ICM B05D005-06  
 INCL 427064000  
 CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)  
 Section cross-reference(s): 76  
 IT 50926-11-9, Indium tin oxide  
 RL: DEV (Device component use); USES (Uses)  
 (electrode; organic/polymer electroluminescent device fabrication)  
 IT 193-44-2, Tetrathiatetracene 198-55-0, Perylene. 1518-16-7, 7,7,8,8,-Tetracyano-p-quinodimethane 31366-25-3, Tetrathiafulvalene 54627-88-2, 1-Methyl-1,4-dithianium 55259-49-9, Tetramethyltetraselenafulvalene 62025-91-6D, metal compds. with tetra-n-butylammonium 66946-48-3, Bis(ethylenedithio)-tetrathiafulvalene 98507-06-3 101683-17-4 118148-29-1 120120-58-3  
 RL: DEV (Device component use); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses)  
 (organic/polymer electroluminescent device fabrication)  
 REFERENCE COUNT: 3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L55 ANSWER 23 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN  
 ACCESSION NUMBER: 1998:415748 HCAPLUS Full-text  
 DOCUMENT NUMBER: 129:167225  
 TITLE: Effect of **electrode**-materials for electrocrystallization of organic charge-transfer complex (TMTSF)<sub>2</sub>ClO<sub>4</sub>  
 AUTHOR(S): Anzai, Hiroyuki; Maki, Suguru; Takasaki, Satoshi; Tanaka, Satoru; Nakatsuji, Shin'ichi; Yamada, Jun-ichi; Nozaki, Ken; Negishi, Akira; Harusawa, Miho  
 CORPORATE SOURCE: Ako-gun, Kanaji Kamigori-cho, 1479-1, Faculty of Science, Department of Material Science, Himeji Institute of Technology, Hyogo, 678-1297, Japan  
 SOURCE: Journal of Crystal Growth (1998), 191(1/2), 148-152  
 CODEN: JCRGAE; ISSN: 0022-0248  
 PUBLISHER: Elsevier Science B.V.  
 DOCUMENT TYPE: Journal  
 LANGUAGE: English  
 AB Several materials (Pt, Au, Pd, Ag, Cu, Ni, glassy carbon, PbO<sub>2</sub>, TiO<sub>2</sub> and SnO<sub>2</sub>-Sb<sub>2</sub>O<sub>3</sub>) as pos. **electrode** and Pt metal as neg. **electrode**, resp., were studied for crystal growth by electrocrystn., to obtain good-quality crystals of (TMTSF)<sub>2</sub>ClO<sub>4</sub> and to reduce the cost of growing crystals.  
 IT 55259-49-9, Tetramethyltetraselenafulvalene  
 RL: PRP (Properties); RCT (Reactant); RACT (Reactant or reagent)  
 (electrochem. oxidation of tetramethyltetraselenafulvalene on various **electrodes** in solution containing tetrabutylammonium perchlorate in electrocrystn.)  
 RN 55259-49-9 HCAPLUS  
 CN 1,3-Diselenole, 2-(4,5-dimethyl-1,3-diselenol-2-ylidene)-4,5-dimethyl- (CA INDEX NAME)



- CC 72-2 (Electrochemistry)  
Section cross-reference(s): 75
- ST **electrode** electrocrystn org chargé transfer complex; TMTSF perchlorate electrocrystn **electrode** material effect; tetramethyltetraselenafulvalenium perchlorate electrocrystn **electrode** material; metal **electrode** TMTSF perchlorate electrocrystn; semiconductor **electrode** TMTSF perchlorate electrocrystn
- IT Metals, uses  
RL: DEV (Device component use); PRP (Properties); USES (Uses)  
(effect of **electrode**-materials for electrocrystn. of organic charge-transfer complex (TMTSF)2ClO4)
- IT Charge transfer complexes  
RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process)  
(effect of **electrode**-materials for electrocrystn. of organic charge-transfer complex (TMTSF)2ClO4)
- IT Crystal growth  
(electrochem.; **electrode**-materials for (TMTSF)2ClO4)
- IT Crystallization  
(electrocrystallization; effect of **electrode**-materials for electrocrystn. of organic charge-transfer complex (TMTSF)2ClO4)
- IT Semiconductor **devices**  
Semiconductor **devices**  
(**electrodes**; for electrocrystn. of organic charge-transfer complex (TMTSF)2ClO4)
- IT **Anodes**  
(for electrocrystn. of organic charge-transfer complex (TMTSF)2ClO4)
- IT Oxidation, electrochemical  
(of tetramethyltetraselenafulvalene on various **electrodes** in solution containing tetrabutylammonium perchlorate in electrocrystn.)
- IT **Electrodes**  
**Electrodes**  
(semiconductive; for electrocrystn. of organic charge-transfer complex (TMTSF)2ClO4)
- IT 7440-22-4, Silver, uses 7440-50-8, Copper, uses  
RL: DEV (Device component use); PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process); USES (Uses)  
(effect of **electrode**-materials for electrocrystn. of organic charge-transfer complex (TMTSF)2ClO4)
- IT 1309-60-0, Lead oxide (PbO2) 7440-05-3, Palladium, uses 7440-06-4, Platinum, uses 7440-44-0, Carbon, uses 7440-57-5, Gold, uses 13463-67-7, Titanium oxide (TiO2), uses 211228-23-8, Antimony tin oxide (Sb0.1Sn0.95O2.05)  
RL: DEV (Device component use); PRP (Properties); USES (Uses)  
(effect of **electrode**-materials for electrocrystn. of organic charge-transfer complex (TMTSF)2ClO4)
- IT 77273-54-2, Bis(tetramethyltetraselenafulvalenium) perchlorate  
RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process)  
(effect of **electrode**-materials for electrocrystn. of



organic charge-transfer complex (TMTSF)2ClO4)

IT 1923-70-2, Tetrabutylammonium perchlorate  
 RL: NUU (Other use, unclassified); PRP (Properties); RCT (Reactant);  
 RACT (Reactant or reagent); USES (Uses)  
 (electrochem. oxidation of tetramethyltetraselenafulvalene on  
 various **electrodes** in solution containing tetrabutylammonium  
 perchlorate in electrocrystn.)

IT 55259-49-9, Tetramethyltetraselenafulvalene  
 RL: PRP (Properties); RCT (Reactant); RACT (Reactant or reagent)  
 (electrochem. oxidation of tetramethyltetraselenafulvalene on  
 various **electrodes** in solution containing tetrabutylammonium  
 perchlorate in electrocrystn.)

IT 7440-02-0, Nickel, uses  
 RL: DEV (Device component use); PRP (Properties); USES (Uses)  
 (**electrode**-materials for attempted electrocrystn. of  
 organic charge-transfer complex (TMTSF)2ClO4)

REFERENCE COUNT: 6 THERE ARE 6 CITED REFERENCES AVAILABLE FOR  
 THIS RECORD. ALL CITATIONS AVAILABLE IN  
 THE RE FORMAT

L55 ANSWER 24 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN  
 ACCESSION NUMBER: 1995:604461 HCAPLUS Full-text  
 DOCUMENT NUMBER: 123:22404  
 TITLE: Active **devices** comprising  
 ferroelectric substances for display  
**devices**  
 INVENTOR(S): Shimada, Shinji  
 PATENT ASSIGNEE(S): Sharp Kk, Japan  
 SOURCE: Jpn. Kokai Tokkyo Koho, 5 pp.  
 CODEN: JKXXAF  
 DOCUMENT TYPE: Patent  
 LANGUAGE: Japanese  
 FAMILY ACC. NUM. COUNT: 1  
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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JP 07084281	A	19950331	JP 1993-229135	199309 14

PRIORITY APPLN. INFO.: JP 1993-229135  
 199309  
 14

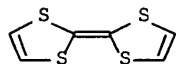
AB The active **devices** comprise an active layer of ferroelec. substances supported  
 between a pair of **electrodes** and the ferroelec. substances contain substances  
 which increase elec. conductivity of the ferroelec. substances. The ferroelec.  
 substances may contain ≥1 selected from poly(vinylidene fluoride), poly(vinyl  
 fluoride), trifluoroethylene, tetrafluoroethylene, and nylon as the monomer unit.  
 The nonlinear active **devices** are used for switching **devices**, e.g. liquid-crystal  
 display **devices**. The **devices** are prevented from separation of the **electrodes** and  
 the active layer. The active layer composed of a film of poly(vinylidene  
 fluoride)-trifluoroethylene copolymer as the 1st ferroelec. layer and a film as  
 the 2nd ferroelec. layer, which comprises the same copolymer as the 1st layer and  
 contains TCNQ and TTF, is exemplified.

IT 31366-25-3, TTF  
 RL: DEV (Device component use); MOA (Modifier or additive use); USES  
 (Uses)  
 (active **devices** comprising active layer of ferroelec.

substances containing additives to increase the elec. conductivity for preventing separation between **electrodes** and the active layer)

RN 31366-25-3 HCAPLUS

CN 1,3-Dithiole, 2-(1,3-dithiol-2-ylidene)- (CA INDEX NAME)



IC ICM G02F001-135

CC 74-13 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

ST active **device** ferroelec substance additive; display active **device** ferroelec substance

IT Ferroelectric substances

Optical imaging **devices**

(active **devices** comprising active layer of ferroelec. substances containing additives to increase the elec. conductivity for preventing separation between **electrodes** and the active layer)

IT Fluoropolymers

RL: DEV (Device component use); USES (Uses)

(active **devices** comprising active layer of ferroelec. substances containing additives to increase the elec. conductivity for preventing separation between **electrodes** and the active layer)

IT Optical instruments

(switches, active **devices** comprising active layer of ferroelec. substances containing additives to increase the elec. conductivity for preventing separation between **electrodes** and the active layer)

IT 28960-88-5, Vinylidene fluoride-trifluoroethylene copolymer

RL: DEV (Device component use); USES (Uses)

(active **devices** comprising active layer of ferroelec. substances containing additives to increase the elec. conductivity for preventing separation between **electrodes** and the active layer)

IT 1518-16-7, TCNQ 31366-25-3, TTF

RL: DEV (Device component use); MOA (Modifier or additive use); USES (Uses)

(active **devices** comprising active layer of ferroelec. substances containing additives to increase the elec. conductivity for preventing separation between **electrodes** and the active layer)

L55 ANSWER 25 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1993:263541 HCAPLUS Full-text

DOCUMENT NUMBER: 118:263541

TITLE: Electrochromic optical switching **device**

INVENTOR(S): Lampert, Carl M.; Visco, Steven J.

PATENT ASSIGNEE(S): University of California, USA

SOURCE: U.S., 9 pp.

CODEN: USXXAM

DOCUMENT TYPE: Patent

LANGUAGE: English

FAMILY ACC. NUM. COUNT: 3

PATENT INFORMATION:

PATENT NO. -----	KIND ----	DATE -----	APPLICATION NO. -----	DATE
US 5142406	A	19920825	US 1990-606063	199010 30
US 5442478	A	19950815	US 1992-872830	199204 23
PRIORITY APPLN. INFO.:			US 1990-606063	A2 199010 30

AB Electrochromic cells are described which comprise an electrochromic **electrode** coupled via an ion-transporting elec. insulating separator with a counter **electrode** formed from a reversibly polymerizable compound described by the general formula (RSy)<sub>n</sub> in the charged state (y = 1-6; n = 2-1,000,000; and R = ≥1 of the same or different C1-20 aliphatic or aromatic organic moieties which may include ≥1 O, S, or N heteroatoms when R comprises ≥1 aromatic rings or ≥1 O, S, N, or F atoms associated with the chain when R comprises an aliphatic chain, aliphatic chains may be linear, branched, saturated or unsatd., and either aliphatic chains or aromatic rings may have substituents). Electrochromic **devices** employing the cells are also described.

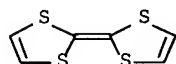
IT 31366-25-3, Tetrathiafulvalene

RL: USES (Uses)

(electrochromic cells with electrochromic **electrodes** from, and organosulfur compound counter **electrodes**)

RN 31366-25-3 HCAPLUS

CN 1,3-Dithiole, 2-(1,3-dithiol-2-ylidene)- (CA INDEX NAME)



IC ICM G02F001-153

INCL 359269000

CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

Section cross-reference(s): 29

ST organosulfur compd **electrode** electrochromic cell;  
reversible polymn **electrode** electrochromic cell

IT Electric contacts

(for electrochromic **devices**, organosulfur compound-containing)

IT Optical imaging **devices**

(electrochromic, with organosulfur compds. counter **electrodes**)

IT Ladder polymers

RL: USES (Uses)

(phenothiazines, electrochromic cells with electrochromic **electrodes** from, and organosulfur compound counter **electrodes**)

IT 1072-71-5, 2,5-Dimercapto-1,3,4-thiadiazole

RL: USES (Uses)

(electrochromic cells with counter **electrodes** containing)

IT 61-73-4, Methylene blue 84-47-9, 2-tert-Butylanthraquinone

84-65-1, Anthraquinone 95-53-4, properties 110-86-1, Pyridine, properties 119-93-7, 4,4'-Diamino-3,3'-dimethylbiphenyl 12030-48-7, Iridium monoxide 12030-49-8, Iridium dioxide 12036-35-0, Rhodium oxide (Rh<sub>2</sub>O<sub>3</sub>) 12054-48-7, Nickel hydroxide (Ni(OH)<sub>2</sub>) 12137-18-7, Rhodium monoxide 13463-67-7, Titanium dioxide, uses 13601-18-8D, solid solution with ferric ferrocyanide 14038-43-8, Ferric ferrocyanide (Fe<sub>4</sub>(Fe(CN)<sub>6</sub>)<sub>3</sub>) 14038-43-8D, solid solution with lithium ferrocyanide 15546-75-5, 5,10-Dihydro-5,10-dimethylphenazine 18933-05-6, Manganese hydroxide (Mn(OH)<sub>2</sub>) 25233-30-1, Polyaniline 25233-34-5, Polythiophene 31366-25-3, Tetrathiafulvalene 36118-45-3, Pyrroline 36490-78-5 46040-54-4 54968-01-3, Iridium hydroxide (Ir(OH)<sub>3</sub>) 56321-86-9, Ruthenium hydroxide 59458-40-1, Gold tungsten oxide 79079-35-9 101178-33-0 116066-80-9, Osmium hydroxide 142448-10-0, Rhodium hydroxide 147657-45-2, Platinum tungsten oxide

RL: USES (Uses)

(electrochromic cells with electrochromic **electrodes** from, and organosulfur compound counter **electrodes**)

IT 1304-76-3, Bismuth oxide (Bi<sub>2</sub>O<sub>3</sub>), properties 1307-96-6, Cobalt monoxide, properties 1308-38-9, Chromium oxide (Cr<sub>2</sub>O<sub>3</sub>), properties 1309-60-0, Lead dioxide 1313-27-5, Molybdenum trioxide, properties 1313-96-8, Niobium oxide (Nb<sub>2</sub>O<sub>5</sub>) 1313-99-1, Nickel monoxide, properties 1314-35-8, Tungsten trioxide, properties 1314-62-1, Vanadium oxide (V<sub>2</sub>O<sub>5</sub>), properties 1317-36-8, Lead monoxide, properties 1317-38-0, Copper oxide (CuO), properties 1343-93-7 1344-43-0, Manganese monoxide, properties 1344-54-3, Titanium oxide (Ti<sub>2</sub>O<sub>3</sub>) 6159-05-3

RL: PRP (Properties)

(electrochromic cells with electrochromic **electrodes** from, and organosulfur compound counter **electrodes**)

IT 7440-74-6, Indium, uses

RL: USES (Uses)

(electrochromic cells with **electrodes** based on zinc monoxide doped with, and organosulfur compds. counter **electrodes**)

IT 18282-10-5, Tin dioxide

RL: USES (Uses)

(electrochromic cells with **electrodes** based on, and organosulfur counter **electrodes**)

IT 7440-36-0, Antimony, uses 7782-41-4, Fluorine, uses

RL: USES (Uses)

(electrochromic cells with **electrodes** from tin oxide doped with, and organosulfur compound counter **electrodes**)

IT 1312-43-2, Indium oxide (In<sub>2</sub>O<sub>3</sub>) 7429-90-5, Aluminum, uses 7440-02-0, Nickel, uses 7440-16-6, Rhodium, uses 7440-22-4, Silver, uses 7440-32-6, Titanium, uses 7440-47-3, Chromium, uses 7440-50-8, Copper, uses 7440-57-5, Gold, uses 12014-13-0, Cadmium tin oxide (CdSnO<sub>3</sub>) 12185-56-7, Cadmium stannate (Cd<sub>2</sub>SnO<sub>4</sub>) 12597-68-1, Stainless steel, properties 12597-71-6, Brass, uses 22205-45-4, Copper sulfide (Cu<sub>2</sub>S) 25583-20-4, Titanium mononitride 37271-26-4, Titanium oxynitride

RL: USES (Uses)

(electrochromic cells with **electrodes** from, and organosulfur compound counter **electrode**)

IT 1306-19-0, Cadmium monoxide, properties 1314-13-2, Zinc monoxide, properties

RL: PRP (Properties)

(electrochromic cells with **electrodes** from, and organosulfur compound counter **electrode**)

IT 33454-82-9, Lithium triflate

RL: USES (Uses)

(electrochromic **devices** with layers containing, with  
organosulfur compound counter **electrodes**)

L55 ANSWER 26 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1990:568254 HCAPLUS Full-text

DOCUMENT NUMBER: 113:168254

TITLE: Comparative study of first-, second- and  
third-generation amperometric glucose enzyme  
**electrodes** in continuous-flow analysis  
of undiluted whole blood

AUTHOR(S): Gunasingham, Hari; Tan, Chin Huat; Aw, Tar Choon

CORPORATE SOURCE: Dep. Chem., Natl. Univ. Singapore, Kent Ridge,  
0511, Singapore

SOURCE: Analytica Chimica Acta (1990), 234(2), 321-30

CODEN: ACACAM; ISSN: 0003-2670

DOCUMENT TYPE: Journal

LANGUAGE: English

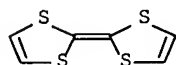
AB First-, second-, and third-generation amperometric glucose enzyme **electrodes** were compared under flow-injection and steady-state conditions for the monitoring of undiluted whole blood. First-generation **electrodes**, based on the detection of hydrogen peroxide at a platinum **electrode**, are generally unsuitable because of the eventual poisoning of the **electrode** and because of their susceptibility to oxygen variation. Second-generation **electrodes** in which a mediator is used for the reoxidn. of glucose oxidase are more suitable for the anal. of whole blood under both steady-state and flow-injection conditions. However, the choice of mediator is important. The best results with regard to linear range and stability were obtained with tetrathiafulvalene, whereas dimethylferrocene required considerable pretreatment before use. A third-generation **electrode** based on tetrathiafulvalene- tetracyanoquinodimethane where direct oxidation of glucose oxidase occurs also proved useful but showed lower stability and a smaller dynamic range compared with the second-generation **devices**. Flow-injection and steady-state studies were carried out using wall-jet cell geometry.

IT 31366-25-3

RL: ANST (Analytical study)  
(in glucose-selective enzyme **electrode**)

RN 31366-25-3 HCAPLUS

CN 1,3-Dithiole, 2-(1,3-dithiol-2-ylidene)- (CA INDEX NAME)



CC 9-1 (Biochemical Methods)

Section cross-reference(s): 72

ST amperometric glucose enzyme **electrode**; blood glucose detn

IT Blood analysis

(glucose determination in, amperometric enzyme **electrodes**  
comparison for)

IT **Electrodes**

(bio-, enzyme, glucose-selective, amperometric, for  
continuous-flow anal. of blood)

IT 50-99-7

RL: ANST (Analytical study)  
(blood analysis, glucose determination in, amperometric enzyme  
**electrodes** comparison for)

IT 50-99-7, Glucose, biological studies

RL: ANT (Analyte); ANST (Analytical study)  
(detn of, amperometric enzyme **electrodes** comparison  
for)

IT 1291-47-0, Dimethylferrocene 1518-16-7 31366-25-3

RL: ANST (Analytical study)  
(in glucose-selective enzyme **electrode**)

L55 ANSWER 27 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1990:524924 HCAPLUS Full-text

DOCUMENT NUMBER: 113:124924

TITLE: Memory devices utilizing fulvalene derivative  
thin films

INVENTOR(S): Sukegawa, Takeshi; Maruno, Toru; Hayashida,  
Shoichi

PATENT ASSIGNEE(S): Nippon Telegraph and Telephone Corp., Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 10 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent

LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. -----	KIND ----	DATE -----	APPLICATION NO. -----	DATE
JP 02060166	A	19900228	JP 1988-210542	198808 26
PRIORITY APPLN. INFO.: JP 1988-210542				198808 26

AB A single memory unit of the title memory device comprises a liquid or solid electrolyte, a working **electrode**, a counter **electrode**, and optionally a reference **electrode**, the working **electrode** bearing a vapor-deposited thin film of a chalcogen-containing fulvalene derivative. A single unit may also comprise a liquid or solid electrolyte and 2 sets of **electrodes** where 1 working **electrode** bears a thin film of a chalcogen-containing vapor-deposited fulvalene derivative and the other working **electrode** bears a thin film of an organic compound or organometallic complex having a reversible oxidation-reduction potential between the oxidation and reduction potentials of the deposited fulvalene film. The deposited films of the organic compound or organometallic complex and the fulvalene derivative are in contact with each other. The fulvalene films have oxidation and reduction potentials which differ in (absolute) value, extremely rapid oxidation and reduction reaction rates, are elec. conductive in the oxidized state, and allow switching between the potentials for the oxidation and reduction reactions. Highly integrated fast response devices can be obtained.

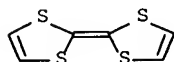
IT 31366-25-3, Tetrathiafulvalene 55259-49-9,  
Tetramethyltetraselenafulvalene 66946-48-3,  
Bisethylenedithiotetrathiafulvalene

RL: USES (Uses)

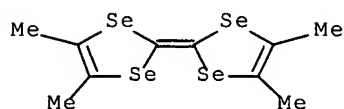
(**electrochem.** memory devices using)

RN 31366-25-3 HCAPLUS

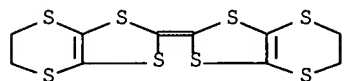
CN 1,3-Dithiole, 2-(1,3-dithiol-2-ylidene)- (CA INDEX NAME)



RN 55259-49-9 HCAPLUS  
 CN 1,3-Diselenole, 2-(4,5-dimethyl-1,3-diselenol-2-ylidene)-4,5-dimethyl- (CA INDEX NAME)



RN 66946-48-3 HCAPLUS  
 CN 1,3-Dithiolo[4,5-b][1,4]dithiin, 2-(5,6-dihydro-1,3-dithiolo[4,5-b][1,4]dithiin-2-ylidene)-5,6-dihydro- (CA INDEX NAME)



IC ICM H01L029-28  
 ICS H01L027-10  
 CC 76-14 (Electric Phenomena)  
 Section cross-reference(s): 72, 74  
 ST memory **device** fulvalene film **electrochem**  
 IT Memory **devices**  
 (electrochem., fulvalene derivs. for)  
 IT 102-54-5, Ferrocene **31366-25-3**, Tetrathiafulvalene  
**55259-49-9**, Tetramethyltetraselenafulvalene  
**66946-48-3**, Bisethylenedithiotetrathiafulvalene  
 RL: **USES (Uses)**  
 (electrochem. memory **devices** using)

L55 ANSWER 28 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN  
 ACCESSION NUMBER: 1990:130820 HCAPLUS Full-text  
 DOCUMENT NUMBER: 112:130820  
 TITLE: Switching device  
 INVENTOR(S): Eguchi, Takeshi; Kawada, Harunori; Sakai,  
 Kunihiro; Matsuda, Hiroshi  
 PATENT ASSIGNEE(S): Canon K. K., Japan  
 SOURCE: Jpn. Kokai Tokkyo Koho, 13 pp.  
 CODEN: JKXXAF  
 DOCUMENT TYPE: Patent  
 LANGUAGE: Japanese  
 FAMILY ACC. NUM. COUNT: 1  
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
-----	----	-----	-----	
JP 01245577	A	19890929	JP 1988-71762	198803 28
PRIORITY APPLN. INFO.:			JP 1988-71762	198803

28

AB A stable switching device with an improved reproducibility comprises an organic insulator layer having a periodic layer structure between a pair of **electrodes**  $\geq 1$  of which comprises an organic conductor.

IT 101853-37-6

RL: USES (Uses)

(elec. switches containing)

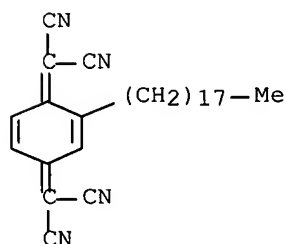
RN 101853-37-6 HCAPLUS

CN Propanedinitrile, 2,2'-(2-octadecyl-2,5-cyclohexadiene-1,4-diylidene)bis-, compd. with 2-(4,5-dimethyl-1,3-dithiol-2-ylidene)-4,5-dimethyl-1,3-dithiole (1:1) (9CI) (CA INDEX NAME)

CM 1

CRN 101853-36-5

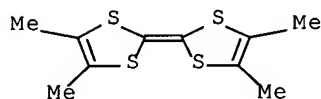
CMF C30 H40 N4



CM 2

CRN 50708-37-7

CMF C10 H12 S4



IC ICM H01L049-02

ICS H01L029-28

ICA C08G061-00; C08G073-00

CC 76-14 (Electric Phenomena)

ST switch org insulator **electrode**

IT Electric switches and switching

(organic **electrode** and insulators for)

IT 110-00-9D, derivs., polymers 9033-83-4, Poly(phenylene)

12369-74-3, Lutetium diphthalocyanine 25014-15-7,

Poly(2-vinylpyridine) 25067-58-7, **Polyacetylene**

25067-59-8 25067-97-4 25135-12-0, Poly(1-vinyl naphthalene)

25135-16-4, Polynaphthalene 25190-62-9, Poly p-phenylene

25212-74-2, Poly p-phenylene sulfide 25233-30-1, Polyaniline

25667-40-7, Poly p-phenylene oxide 26009-24-5, Poly p-phenylene

vinylene 26499-97-8, Poly m-phenylene 27880-39-3,



Poly(1,4-phenylenemethylene) 27987-87-7, Polydiacetylene  
 28406-56-6, Poly(2-vinylnaphthalene) 30604-81-0, Polypyrrole  
 34801-99-5, Poly(vinyl ferrocene) 51325-05-4, Polythienylene  
 52410-66-9, Poly(seleno-1,4-phenylene) 89231-09-4, Polyselenophene  
 91201-85-3 101853-37-6 101909-00-6 112261-44-6

RL: USES (Uses)  
 (elec. switches containing)

L55 ANSWER 29 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1988:104856 HCAPLUS Full-text

DOCUMENT NUMBER: 108:104856

TITLE: Organic-thin-film electric elements

INVENTOR(S): Mizushima, Koichi; Nakayama, Toshio; Miura,  
 Akira; Motoma, Nobuhiro

PATENT ASSIGNEE(S): Toshiba Corp., Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 9 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent

LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. -----	KIND ----	DATE -----	APPLICATION NO. -----	DATE
JP 62222669	A	19870930	JP 1986-66277	198603 25

PRIORITY APPLN. INFO.: JP 1986-66277

198603  
25

AB The title element, used for an electronic **device**, consists of a laminate of alternately placed 1st (donor)- and 2nd (acceptor)-type organic thin films, where part of the **electrode** to apply elec. potential to the laminate is formed of an organic elec. conductor. The films and the elec. conductor may be produced by Langmuir-Blodgett method. The element has improved charge-carrying efficiency when elec. potential is applied.

IT 51159-15-0

RL: USES (Uses)

(organic elec.-conductor **electrodes** from, thin-film elec.  
 elements containing)

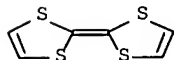
RN 51159-15-0 HCAPLUS

CN Propanedinitrile, 2,2'-(2,5-cyclohexadiene-1,4-diylidene)bis-,  
 compd. with 2-(1,3-dithiol-2-ylidene)-1,3-dithiole (9CI) (CA INDEX  
 NAME)

CM 1

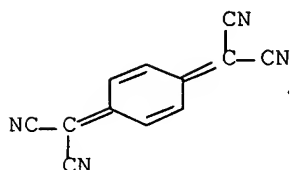
CRN 31366-25-3

CMF C6 H4 S4



CM 2

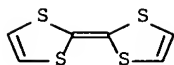
CRN 1518-16-7  
CMF C12 H4 N4



IC ICM H01L029-28  
ICS H01L029-46  
CC 76-2 (Electric Phenomena)  
IT **Electrodes**  
(organic elec.-conductor, thin-film elec. elements containing)  
IT Electric conductors  
(organic, **electrodes** from, thin-film elec. elements containing)  
IT **51159-15-0**  
RL: USES (Uses)  
(organic elec.-conductor **electrodes** from, thin-film elec. elements containing)

L55 ANSWER 30 OF 30 HCAPLUS COPYRIGHT 2007 ACS on STN  
ACCESSION NUMBER: 1977:574763 HCAPLUS Full-text  
DOCUMENT NUMBER: 87:174763  
TITLE: Electrochemical properties of dopants and the d.c. dynamic scattering of a nematic liquid crystal  
AUTHOR(S): Lim, H. S.; Margerum, J. D.; Graube, A.  
CORPORATE SOURCE: Hughes Res. Lab., Malibu, CA, USA  
SOURCE: Journal of the Electrochemical Society (1977), 124(9), 1389-94  
CODEN: JESOAN; ISSN: 0013-4651  
DOCUMENT TYPE: Journal  
LANGUAGE: English

AB Flow of liquid from 1 **electrode** to the other was observed during dynamic scattering of a phenyl benzoate nematic liquid crystal. The direction of the flow depended upon the electrochem. properties of dopants. The flow was from **cathode** to **anode** when the dopant was an electron acceptor, and vice versa when the dopant was a donor. A redox dopant gave distinctively different d.c. dynamic scattering patterns from a salt dopant, and did not give the Williams domain pattern which was observed with a salt dopant. Charge conduction mechanisms through the liquid crystal are discussed in terms of the **electrode** reactions of the liquid crystal components and the dopants.  
IT **31366-25-3**  
RL: PRP (Properties)  
(liquid crystal flow to **electrode** during dynamic scattering in relation to)  
RN 31366-25-3 HCAPLUS  
CN 1,3-Dithiole, 2-(1,3-dithiol-2-ylidene)- (CA INDEX NAME)



- CC 72-11 (Electrochemistry)  
 Section cross-reference(s): 74, 75
- ST liq crystal **electrode** reaction; dopant electrochem  
 property; dynamic scattering liq crystal; flow liq crystal  
**electrode**; phenyl benzoate liq crystal; optical display  
**device**
- IT Optical display **devices**  
 (**electrochem.** properties of dopants in relation to)
- IT **Electrode** reaction  
 (of liquid crystals and dopants, in nonaq. solvents)
- IT 7439-97-6, uses and miscellaneous 7440-06-4, uses and  
 miscellaneous  
 RL: USES (Uses)  
 (**cathode**, phenylbenzoate liquid crystal reduction on, in  
 nonaq. solvent)
- IT 38454-23-8 38454-24-9 52709-88-3 60127-45-9  
 RL: PRP (Properties)  
 (dynamic scattering of, dopant effect on flow to  
**electrode** in relation to)
- IT 1172-07-2 1274-08-4 1518-16-7 1923-70-2 31366-25-3  
 35895-70-6  
 RL: PRP (Properties)  
 (liquid crystal flow to **electrode** during dynamic  
 scattering in relation to)

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